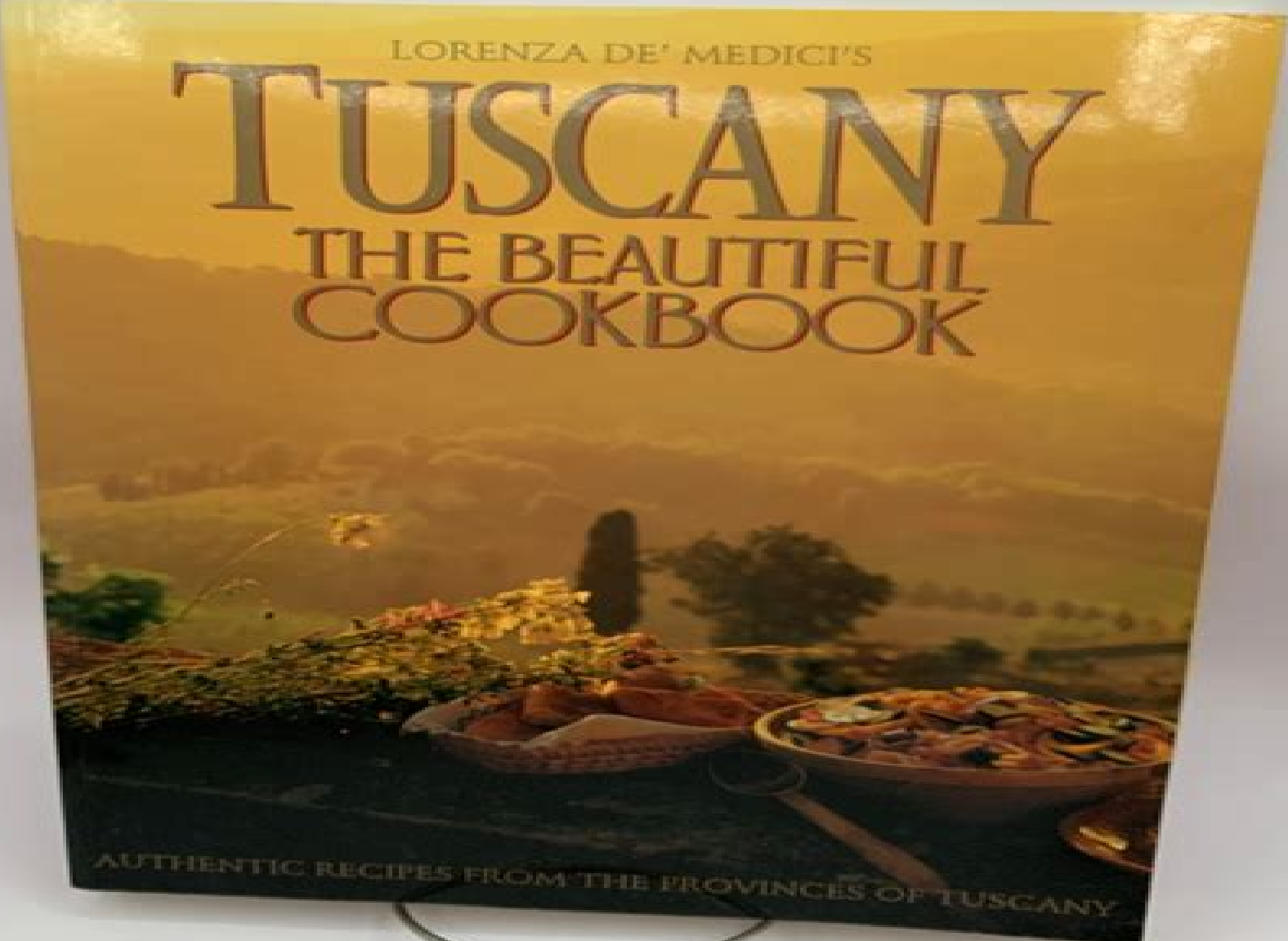


LORENZA DE' MEDICI'S

TUSCANY

THE BEAUTIFUL COOKBOOK



A painting of a Tuscan landscape with rolling hills, a cypress tree, and a basket of bread and a bowl of soup in the foreground.

AUTHENTIC RECIPES FROM THE PROVINCES OF TUSCANY

Tuscany The Beautiful Cookbook

**Lorenza De' Medici, Peter
Johnson, Michael Freeman**



Tuscany The Beautiful Cookbook:

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Art Culinaire ,1998 *The Edible Italian Garden* Rosalind Creasy,1999 Creasy offers recommendations for selecting and growing the best varieties of Italian herbs and vegetables basil broccoli rabe fava beans and tomatoes Over 90 color illustrations Tuscan-American Kitchen, A Cassandra Vivian,Vivian Pellni Sansone,2011-05-04 This definitive Italian cookbook presents more than 250 kitchen tested recipes along with five essays and illustrated step by step instructions for the essentials of Italian cooking such as stuffing an artichoke making cannelloni identifying dozens of types of pasta and more In addition to the well known pastas breads and meats the recipes include canapes sweets baccal chicken galatine soups vegetables porchetta sausage salami and Other cured meats Indexes and recipe listings in both Italian and English and a complete chapter devoted to the gran fritto misto the Italian method for frying meats and vegetables complete this wholesome and hearty celebration of homemade Tuscan cooking *New York* ,1992-10 **The publishers weekly** ,1992

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 $2 \text{ CO}_2 \text{ g} + \text{H}_2\text{O g} + \text{C}_2\text{H}_2 \text{ g} \rightarrow \text{O}_2 \text{ g} + \text{C}_2\text{H}_6 \text{ g} \quad \Delta h = 94.5 \text{ kJ}$
 $\text{H}_2\text{O g} + \text{H}_2 \text{ g} + \frac{1}{2} \text{O}_2 \text{ g} \rightarrow \text{C}_2\text{H}_6 \text{ g} \quad \Delta h = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6 \text{ g} + 3 \text{ H}_2 \text{ O g} \rightarrow \Delta h = 283 \text{ kJ}$
 $\text{C}_2\text{H}_6 \text{ g}$

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 $\text{C}_2\text{H}_4 \text{ g} + 3 \text{ O}_2 \text{ g} \rightarrow 2 \text{ CO}_2 \text{ g} + 2 \text{ H}_2\text{O l} \quad \Delta h = 1411 \text{ kJ}$
 $\text{C}_2\text{H}_6 \text{ g} + 7 \text{ O}_2 \text{ g} \rightarrow 2 \text{ CO}_2 \text{ g} + 3 \text{ H}_2\text{O l} \quad \Delta h = 1560 \text{ kJ}$
 $\text{H}_2 \text{ g} + \frac{1}{2} \text{O}_2 \text{ g} \rightarrow \text{H}_2\text{O l} \quad \Delta h = 285 \text{ kJ}$
 mole

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