

## HEAT

1. Thermal energy **ALWAYS** moves from warmer to cooler.  
a. True      b. False
2. A material that doesn't transfer heat very well is called a(n) \_\_\_\_\_.  
a. applicator      b. protector      c. insulator      d. conveyor
3. Which of the following is not a form of heat transfer?  
a. conduction      b. convection      c. transpiration      d. radiation
4. The transfer of heat by the movement of a mass or substance.  
a. Conduction      b. convection c. Radiation
5. The transfer of energy (heat) through space by electromagnetic waves.  
a. conduction      b. convection      c. radiation
6. Thermal energy can be transferred by conduction from one piece of matter to another only if the two pieces are  
a. liquids      b. touching      c. solids      d. fuel
7. The measure of how hot or cold something is is called:  
a. Density      b. temperature      c. volume      d. solid
8. Which of the following temperatures is the same as absolute zero?  
a. 0 degrees Centigrade      b. -273.15 degrees Fahrenheit      c. 0 Kelvin      d. 0 degrees Fahrenheit
9. To heat 1g of water by 1 degree Celsius requires:  
a. 1 calorie      b. 1 Calorie      c. 1 joule      d. 1 watt
10. As water begins to boil in a kettle the hotter water at the bottom of the kettle closest to the stove begins to rise and the cooler water above sinks and warms. This is an example of:  
a. conduction      b. radiation      c. convection d. heat
11. The room becomes warm when the furnace is turned on: \_\_\_\_\_.  
a. conduction      b. convection c. radiation
12. Bacon cooking in the frying pan is a form of:  
a. conduction b. convection      c. radiation
13. When you add hot water to cold water, the cold water warms up because of  
a. insulation      b. convection c. radiation      d. solar energy
14. The first law of thermodynamics states:  
a. components of a system interact  
b. entropy of a closed system always increases  
c. a closed system always shifts toward disorder or homogeneity  
d. energy is always conserved
15. The process by which temperature drops as a gas expands, without the addition of heat, is called \_\_\_\_\_ cooling.  
a. adiabatic      b. thermodynamic      c. spontaneous      d. cyclonic
16. Why does a piece of steel heat up more than an equal mass of plastic when both absorb the same energy?

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**C Cleary**

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