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web we can write a balanced nuclear equation for this reaction as let s do a practice example identify the product x by

writing a balanced nuclear equation  $^{137}\text{Cs} + ^{55}\text{Cs} \rightarrow ^{137}\text{Ba} + \text{X}$  because the mass number is the same on both sides of the equation  $\text{X}$  must have a

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web identify the missing particle in the following nuclear reaction  $^{10}\text{N} + ^{235}\text{U} \rightarrow ^{137}\text{Cs} + ^{92}\text{Sr} + ^{40}\text{K}$

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