



# States of Matter: Phase Changes

Phase	Description	Illustration
<b>Evaporation</b>	<b>Liquid → _____</b> <ul style="list-style-type: none"><li>❖ Water changes into a gas called _____</li><li>❖ Heat energy is added</li><li>❖ Particles move faster and faster</li><li>❖ Examples include a _____ disappearing and clothes drying</li></ul>	
<b>Condensation</b>	<b>Gas → _____</b> <ul style="list-style-type: none"><li>❖ Water vapor turns into droplets of (liquid) _____</li><li>❖ Heat energy is removed</li><li>❖ Particles _____</li><li>❖ Examples include morning dew and a foggy mirror after a hot shower</li></ul>	
<b>Freezing</b>	<b>Liquid → _____</b> <ul style="list-style-type: none"><li>❖ Heat energy is _____</li><li>❖ (Liquid) water turns into ice</li><li>❖ Particles slow down</li><li>❖ Examples include rain turning into _____ and juice freezing into a popsicle</li></ul>	
<b>Melting</b>	<b>Solid → _____</b> <ul style="list-style-type: none"><li>❖ Heat energy is _____</li><li>❖ Ice turns into (liquid) water</li><li>❖ Particles move _____</li><li>❖ Examples include a snowman thawing out and wax dripping down a candle</li></ul>	

❖ Matter can change from one phase to another with the addition or removal of \_\_\_\_\_

❖ Matter changing from one state to another is a \_\_\_\_\_

# Study Guide Phase Changes

**David D. Reid**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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