

Chapter 5 Study Guide

In your textbook, read about the wave nature of light.

Study your notes, charts, tables, and problems that we have completed in class.
Review information in your Science Workbook

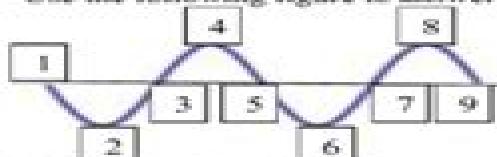
Use the following terms below once to complete the passage.

Amplitude	Energy	Frequency	Hertz
Light	Wave	Wavelength	Speed

Electromagnetic radiation is a kind of (1) Energy that behaves like a(n) (2) Wave as it travels through space. (3) Light is one type of electromagnetic radiation. Other examples include x-rays, radio waves and microwaves.

All waves can be characterized by their wavelength, frequency and (4) Speed. The shortest distance between equivalent points on a continuous wave is called a(n) (5) Wavelength. The height of a wave from the origin to a crest or from the origin to the trough is the (6) Amplitude. (7) Frequency is the number of waves that pass a given point in one second. The SI unit for frequency is the (8) Hertz, which is equivalent to one wave per second.

Use the following figure to answer questions 9-10



9. Which number(s) represents one wavelength? any of the following: 2-6, 4-8, 1-5, 5-9

10. Which number(s) represent the amplitude? any of the following: 1-2, 3-4, 4-5, 6-5, ect.

11. Circle each of the following things that helped scientist discover the location of the electrons. There may be more than one correct answer.

- a. Different elements have different emission spectra.
- b. deBroglie equation predicted that all moving particles have wavelike characteristics.
- c. The Heisenberg Uncertainty Principle
- d. Schrödinger's Wave equation.

12. The minimum amount of energy that can be gained or lost by an atom is called Quantum.

13. What was the major contribution to the understanding of the atom:

Einstein: proposed that light has a dual nature (particle and wave). Beam of light made of bundles of energy called photons. Photons with energy above the threshold will cause the photoelectric effect.

Planck: determined that matter gains or loses energy in small amounts called quantum.

Study Guide Light And Quantized Answer

Patricia L. Samuel

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Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone,Theodore O. Dolter,2012-01-03

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