

CHAPTER 14 STUDY GUIDE

Mixtures and Solutions

Section 14.1 Heterogeneous and Homogeneous Mixtures

In your textbook, read about suspensions and colloids.

For each statement below, write *true* or *false*.

- | | |
|--------------|---|
| <u>false</u> | 1. A solution is a mixture containing particles that settle out of the mixture if left undisturbed. |
| <u>true</u> | 2. The most abundant substance in a colloid is the dispersion medium. |
| <u>false</u> | 3. A colloid can be separated by filtration. |
| <u>true</u> | 4. A solid emulsion consists of a liquid dispersed in a solid. |
| <u>true</u> | 5. Whipped cream is an example of a foam. |
| <u>false</u> | 6. In an aerosol, the dispersing medium is a liquid. |
| <u>true</u> | 7. Brownian motion results from the collisions of particles of the dispersion medium with the dispersed particles. |
| <u>true</u> | 8. Dispersed particles in a colloid do not tend to settle out because they have polar or charged atomic groups on their surfaces. |
| <u>false</u> | 9. Stirring an electrolyte into a colloid stabilizes the colloid. |
| <u>true</u> | 10. Colloids demonstrate the Tyndall effect. |

The table below lists the characteristics of particles in colloids, solutions, and suspensions. Place a check in the column of each mixture whose particles have a particular characteristic.

Characteristics of Particles	Colloid	Solution	Suspension
11. Less than 1 nm in diameter		<input checked="" type="checkbox"/>	<input checked="" type="checkbox"/>
12. Between 1 nm and 1000 nm in diameter	<input checked="" type="checkbox"/>	<input checked="" type="checkbox"/>	<input checked="" type="checkbox"/>
13. More than 1000 nm in diameter	<input checked="" type="checkbox"/>	<input checked="" type="checkbox"/>	
14. Settle out if undisturbed			
15. Pass through standard filter paper	<input checked="" type="checkbox"/>		
16. Lower vapor pressure			<input checked="" type="checkbox"/>
17. Scatter light	<input checked="" type="checkbox"/>		

Study Guide Mixtures And Solutions Answer Key

Leo J. Malone



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Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03
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Table of Contents Study Guide Mixtures And Solutions Answer Key

1. Understanding the eBook Study Guide Mixtures And Solutions Answer Key
 - The Rise of Digital Reading Study Guide Mixtures And Solutions Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Study Guide Mixtures And Solutions Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Study Guide Mixtures And Solutions Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Study Guide Mixtures And Solutions Answer Key
 - Personalized Recommendations
 - Study Guide Mixtures And Solutions Answer Key User Reviews and Ratings
 - Study Guide Mixtures And Solutions Answer Key and Bestseller Lists
5. Accessing Study Guide Mixtures And Solutions Answer Key Free and Paid eBooks
 - Study Guide Mixtures And Solutions Answer Key Public Domain eBooks
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 - Study Guide Mixtures And Solutions Answer Key Enhanced eBook Features
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 - Highlighting and Note-Taking Study Guide Mixtures And Solutions Answer Key
 - Interactive Elements Study Guide Mixtures And Solutions Answer Key
8. Staying Engaged with Study Guide Mixtures And Solutions Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Study Guide Mixtures And Solutions Answer Key
9. Balancing eBooks and Physical Books Study Guide Mixtures And Solutions Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Study Guide Mixtures And Solutions Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Study Guide Mixtures And Solutions Answer Key
 - Setting Reading Goals Study Guide Mixtures And Solutions Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Study Guide Mixtures And Solutions Answer Key
 - Fact-Checking eBook Content of Study Guide Mixtures And Solutions Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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