

Practice Test #21: Nuclear Chemistry

1. Of the following processes, which one does not change the atomic number?

A) alpha emission ${}_Z^XA \rightarrow {}_{Z-2}^{A-4}B + {}_2^4\alpha$

B) beta emission ${}_Z^XA \rightarrow {}_{Z+1}^XB + {}_{-1}^0\beta$

C) electron capture ${}_Z^XA + {}_{-1}^0e(\text{orbital electron}) \rightarrow {}_{Z-1}^XB$

D) gamma emission ${}_Z^XA^* \rightarrow {}_Z^XA + {}_0^0\gamma$ (only this nuclear reaction does not change the atomic number)

E) positron emission ${}_Z^XA \rightarrow {}_{Z-1}^XB + {}_{+1}^0e$

2. In balancing the nuclear reaction below, the identity of element "E" is determined from its...



A) atomic weight.

B) mass number.

C) atomic number.

D) number of neutrons.

E) number of electrons.

3. The largest number of stable nuclei have an ____ number of protons and an ____ number of neutrons.

A) even, even

B) odd, odd

C) even, odd

D) odd, even

4. The reaction below is an example of _____ decay.



A) alpha

B) beta

C) gamma

D) positron

Study Guide Nuclear Chemistry Answer Key

Paul B. Kelter



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