

## Section 3.1: Early Atomic Theories and the Origins of Quantum Theory

### Research This: The Large Hadron Collider, page 141

A. String theory is a theory of matter that describes sub-atomic particles as strings. Grand unified theory is an attempt to unite the fundamental forces of nature: electromagnetic force, gravity, weak nuclear force, and strong nuclear force.

B. Answers may vary. Sample answer: The LHC and ATLAS projects are worth the enormous amount of money because the information collected will improve our knowledge of sub-atomic particles and will improve our understanding of the components of the universe.

C. Answers may vary. Students' answers should allow the audience to learn about the basic concepts that are involved in the LHC and ATLAS research projects.

D. Answers may vary. Sample answer: Further investments in these projects and this type of research should be continued because this type of research requires massive collaboration and funding. By funding this type of pure experimentation, scientific knowledge can be advanced in unexpected ways.

### Section 3.1 Questions, page 142

1. (a)(i)  $^{75}\text{Br}$  has 35 protons and 44 neutrons.  
(ii)  $^{81}\text{Br}$  has 35 protons and 46 neutrons.  
(iii)  $^{239}\text{Pu}$  has 94 protons and 145 neutrons.  
(iv)  $^{133}\text{Cs}$  has 55 protons and 78 neutrons.  
(v)  $^3\text{H}$  has 1 proton and 2 neutrons.  
(vi)  $^{56}\text{Fe}$  has 26 protons and 30 neutrons.  
(b)(i) The neutral atom for  $^{75}\text{Br}$  has 35 electrons.  
(ii) The neutral atom for  $^{81}\text{Br}$  has 35 electrons.  
(iii) The neutral atom for  $^{239}\text{Pu}$  has 94 electrons.  
(iv) The neutral atom for  $^{133}\text{Cs}$  has 55 electrons.  
(v) The neutral atom for  $^3\text{H}$  has 1 electron.  
(vi) The neutral atom for  $^{56}\text{Fe}$  has 26 electrons.
2. (a) The atomic symbol for the isotope with  $Z = 8$ , and 9 electrons is  $^{17}\text{O}$ .  
(b) The atomic symbol for the isotope of chlorine in which  $A = 37$  is  $^{37}\text{Cl}$ .  
(c) The atomic symbol for the isotope with  $Z = 27$  and  $A = 60$  is  $^{60}\text{Co}$ .  
(d) The atomic symbol for the isotope with 26 protons and 31 neutrons is  $^{57}\text{Fe}$ .  
(e) The atomic symbol for the isotope of I with a mass number of 131 is  $^{131}\text{I}$ .  
(f) The atomic symbol for the isotope with  $Z = 3$  and 4 neutrons is  $^3\text{Li}$ .
3. (a)  $\text{Ba}^{2+}$  contains 56 protons and 54 electrons.  
(b)  $\text{Zn}^{2+}$  contains 30 protons and 28 electrons.  
(c)  $\text{N}^{3-}$  contains 7 protons and 10 electrons.  
(d)  $\text{Rb}^+$  contains 37 protons and 36 electrons.  
(e)  $\text{Co}^{2+}$  contains 27 protons and 24 electrons.  
(f)  $\text{Tc}^{3-}$  contains 52 protons and 54 electrons.

# Study Guide Physics Quantum Theory Answer Key

**RJ Shavelson**

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## **Table of Contents Study Guide Physics Quantum Theory Answer Key**

1. Understanding the eBook Study Guide Physics Quantum Theory Answer Key
  - The Rise of Digital Reading Study Guide Physics Quantum Theory Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Study Guide Physics Quantum Theory Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Study Guide Physics Quantum Theory Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Study Guide Physics Quantum Theory Answer Key
  - Personalized Recommendations
  - Study Guide Physics Quantum Theory Answer Key User Reviews and Ratings
  - Study Guide Physics Quantum Theory Answer Key and Bestseller Lists
5. Accessing Study Guide Physics Quantum Theory Answer Key Free and Paid eBooks
  - Study Guide Physics Quantum Theory Answer Key Public Domain eBooks
  - Study Guide Physics Quantum Theory Answer Key eBook Subscription Services
  - Study Guide Physics Quantum Theory Answer Key Budget-Friendly Options
6. Navigating Study Guide Physics Quantum Theory Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Study Guide Physics Quantum Theory Answer Key Compatibility with Devices
  - Study Guide Physics Quantum Theory Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Study Guide Physics Quantum Theory Answer Key
  - Highlighting and Note-Taking Study Guide Physics Quantum Theory Answer Key
  - Interactive Elements Study Guide Physics Quantum Theory Answer Key
8. Staying Engaged with Study Guide Physics Quantum Theory Answer Key

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Study Guide Physics Quantum Theory Answer Key

9. Balancing eBooks and Physical Books Study Guide Physics Quantum Theory Answer Key

- Benefits of a Digital Library
- Creating a Diverse Reading Collection Study Guide Physics Quantum Theory Answer Key

10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

11. Cultivating a Reading Routine Study Guide Physics Quantum Theory Answer Key

- Setting Reading Goals Study Guide Physics Quantum Theory Answer Key
- Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Study Guide Physics Quantum Theory Answer Key

- Fact-Checking eBook Content of Study Guide Physics Quantum Theory Answer Key
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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