



Reaction Rates and Equilibrium Study Guide

ANSWER KEY

1. Describe how each of the following factors would affect the rate of a chemical reaction. Use the term "reactant particle collisions" in your response.
 - a. Temperature decreases. *If temperature decreases, the particles will move more slowly. There will be less reactant particle collisions AND a smaller percentage of particles will have enough energy to produce a reaction. The reaction rate will therefore decrease.*
 - b. Addition of iodide ions, I^- , as a catalyst. *The presence of the catalyst increases the rate of the reaction taking place. This means that a larger percentage of particles will have enough energy to produce a reaction. The reaction rate will therefore increase.*
 - c. Increased reactant concentration. *With more reactant particles present, there will be more collisions between those particles. This means the reaction rate will increase.*
 - d. Crushing a solid into a powder (with larger surface area). *As a powder, there is more surface area for reactant particles to come into contact with each other. This means there will be more collisions, leading to a faster reaction rate.*
2. The reaction progress diagram shows the energy changes that take place when nitrogen dioxide reacts with oxygen to form nitrogen monoxide and oxygen monoxide.

Study Guide Reaction Rates And Equilibrium Answers

Susan McMurry

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Chemical Principles Student's Study Guide & Solutions Manual John Krenos, Joseph Potenza, Loretta Jones, Lynn Koplitz, Thomas Spence, 2004-03-19 This combination manual is designed to help students avoid common mistakes and understand the material better. The solutions manual section includes detailed answers and explanations to the odd numbered exercises in the text.

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step by step examples. Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning. This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today Seager/Slabaugh, 2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text. This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises. New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application.

Organic Chemistry, 5e Student Study Guide and Solutions Manual David R. Klein, Laurie S. Starkey, 2025-03-18 Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems. With Organic Chemistry Student Study Guide and Solutions Manual 5th Edition students can learn to become proficient at approaching new situations methodically based on a repertoire of skills. These skills are vital for successful problem solving in organic chemistry.

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Study Guide/Selected Solutions Manual Julie R. Frentrup, 2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book.

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General, Organic, and Biological Chemistry Study Guide and Selected Solutions Karen C. Timberlake, 2001-11
Keyed to the learning goals in the text this guide is designed to promote active learning through a variety of exercises with answers and mastery exams The guide also contains complete solutions to odd numbered problems

Study Guide and Solutions Manual, Fundamentals of General, Organic, and Biological Chemistry, Third Edition John McMurry, Susan McMurry, 1999 Provides worked out solutions to text problems along with chapter by chapter outlines and a variety of self tests at the end of each chapter

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Raymond, 2005-10-07 Finally readers have a shorter less intimidating introduction to general organic and biological chemistry. Not only is Raymond's text concise, it also takes an integrated approach to presenting important topics in a way that makes the material easier to understand. In this approach, similarities can be exploited and concepts reinforced. The result is that readers see the strong connections that exist between these three branches of chemistry. **Organic Chemistry, Study Guide and Solutions Manual** T. W. Graham Solomons, Craig B. Fryhle, 2000 On the cover of this book is a Pacific yew tree found in the ancient forests of the Pacific Northwest. The bark of the Pacific yew tree produces Taxol, found to be a highly effective drug against ovarian and breast cancer. Taxol blocks mitosis during eukaryotic cell division. The supply of Taxol from the Pacific yew tree is vanishingly small; however, a single 100 year old tree provides only about one dose of the drug, roughly 300 mg. For this reason, as well as the spectacular molecular architecture of Taxol, synthetic organic chemists fiercely undertook efforts to synthesize it. Five total syntheses of Taxol have thus far been reported. Now a combination of isolation of a related metabolite from European yew needles and synthesis of Taxol from that intermediate supply the clinical demand. This case clearly demonstrates the importance of synthesis and the use of organic chemistry. It's just one of the many examples used in the text that will spark the interest of students and get them involved in the study of organic chemistry.

Fundamentals of Engineering Study Guide Lloyd T. Cheney, 1987 This is a 2 volume set including a 288 page Study Guide plus a 144 page Solution Manual. The manual was contributed by 10 professors. It is a review of Mathematics, Chemistry, Statics, Fluid Mechanics, Strength of Materials, Thermodynamics, Electric Circuits, Engineering Economy, Computer Science, and Systems, plus 550 EIT problems with detailed solutions. 50% text, 50% Problems and solutions. The two items are shrink wrapped together. **Study Guide with Selected Solutions** Karen Timberlake, 2002-08-06 Written by the author, the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams. Also contains complete solutions to odd numbered problems.

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