

Quantum chemistry Study Questions

Guide

1. What is quantum mechanics used to describe?
 - A) Large objects
 - B) Macroscopic objects
 - C) Microscopic objects
 - D) Astronomical objects
2. What is the Schrödinger equation used for?
 - A) Describing the behavior of macroscopic objects
 - B) Describing the behavior of microscopic objects
 - C) Describing the behavior of celestial bodies
 - D) Describing the behavior of subatomic particles
3. What is the wave-particle duality of quantum mechanics?
 - A) Particles can act as waves and vice versa
 - B) Particles can only act as waves
 - C) Waves can only act as particles
 - D) There is no wave-particle duality
4. What is the Heisenberg uncertainty principle?
 - A) The more precisely the position of a particle is known, the less precisely its momentum can be known
 - B) The more precisely the momentum of a particle is known, the less precisely its position can be known
 - C) The position and momentum of a particle can be known precisely at the same time
 - D) The position and momentum of a particle are unrelated

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