

# Properties of Common Minerals

Location	Hardness	Crystal Fracture	Color/Trans- lucence	Characteristic Chemical Reactions	Streak	Chemical Formula	Abundance, %wt.
Bedrock (soil)	7-8	✓	black to grey	black - strong greenish	greenish black	C	Graphite
	8-9	✓	metallic black	grey-black - strong - rubro-olivaceous density = 7.8 g/cm <sup>3</sup>	grey-black	PbS	Galena
	9-10-10.5	✓	black to grey	black - strong magnetic	grey-black	Fe <sub>3</sub> O <sub>4</sub>	Magnetite
	9-10	✓	brassy yellow	greenish-black - strong (soft) gold	grey-black	Pb <sub>3</sub> Sn <sub>2</sub>	Pyrite
Other	8.5-9.5 or 10	✓	metallic silver or earthy lead	red-brown - strong	grey-black	Pb <sub>3</sub> Sn <sub>2</sub>	Hemattite
Bedrock (soil)	1	✓	white to green	greenish	colorless powder	Al <sub>2</sub> (OH) <sub>4</sub> Si <sub>2</sub> O <sub>5</sub> (OH) <sub>2</sub>	Quartz
	2	✓	yellow to brown	white-yellow - strong	yellow powder	Si	Sulfur
	2	✓	white to grey or grey	weakly scratched - hydrothermal	glassy or earthy powder	CaSiO <sub>3</sub> - 2H <sub>2</sub> O	Calcite - quartz
	2-2.5	✓	colorless to yellow	fracture in thin sheets	glassy - melting	Si <sub>2</sub> (OH) <sub>4</sub> Si <sub>2</sub> O <sub>5</sub> (OH) <sub>2</sub>	Muscovite mica
	2.5	✓	colorless to white	rubro-olivaceous - earthy lead	black - melting	FeS	Pyrite
	2.5-3	✓	black to dark brown	fracture in thin sheets	colorless - melting	Fe <sub>3</sub> (OH) <sub>4</sub> Si <sub>2</sub> O <sub>5</sub> (OH) <sub>2</sub>	Illite-mica
	3	✓	colorless or variable	fracture with weak - hydrothermal cleavage	colorless - strong	CaCO <sub>3</sub>	Calcite
	3.5	✓	colorless or variable	fracture with weak - when powdered	fracture - strong	CaMgCO <sub>3</sub>	Dolomite
	4	✓	colorless or variable	cleavage in 4 directions	hydrothermal - acid	CaF <sub>2</sub>	Fluorite
	5-6	✓	black to dark green	cleavage in 2 directions at 90°	metallic - colorless - strong	Ca <sub>2</sub> (OH) <sub>4</sub> Si <sub>2</sub> O <sub>5</sub> (OH) <sub>2</sub>	Pyroxene - commonly cleaved
	5.5	✓	black to dark green	cleavage at 90° and 120°	metallic - colorless - strong	Ca <sub>2</sub> (OH) <sub>4</sub> Si <sub>2</sub> O <sub>5</sub> (OH) <sub>2</sub>	Amphibole - commonly cleaved
	6	✓	white to grey	cleavage in 2 directions at 90°	colorless - strong	Ca <sub>2</sub> (OH) <sub>4</sub> Si <sub>2</sub> O <sub>5</sub> (OH) <sub>2</sub>	Pyroxene - commonly cleaved
	6	✓	white to grey	cleavage in 2 directions - strong - weak	colorless - strong	Ca <sub>2</sub> (OH) <sub>4</sub> Si <sub>2</sub> O <sub>5</sub> (OH) <sub>2</sub>	Pyroxene - commonly cleaved
	6.5	✓	green to grey - brown	commonly light green - and granular	metallic - colorless - strong	Ca <sub>2</sub> (OH) <sub>4</sub> Si <sub>2</sub> O <sub>5</sub> (OH) <sub>2</sub>	Pyroxene
	7	✓	colorless or variable	glassy - fracture - hydrothermal - hydrothermal - hydrothermal	glassy - strong - colorless	SiO <sub>2</sub>	Quartz
	8.5-9.5	✓	dark red to green	often brown - red - glassy - granular - earthy - metallic - colorless - strong	strong - colorless - strong	Fe <sub>3</sub> (OH) <sub>4</sub> Si <sub>2</sub> O <sub>5</sub> (OH) <sub>2</sub>	Quartz

\*Chemical symbols

Al = aluminum  
C = carbon  
Ca = calcium

Cl = chlorine  
F = fluorine  
Fe = iron

H = hydrogen  
K = potassium  
Mg = magnesium

Na = sodium  
O = oxygen  
Si = silicon

S = sulfur  
Sn = tin  
Ti = titanium

✓ = cleavage - fracture of cleavage

# Study Guide Mineral Identification Answer Key

**Lincoln Ridler Page, Fredrick Christian  
Kruger**



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