



The Mole Study Guide

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Avogadro's number refers to the number of particles in one gram of a substance.

- ☐ True
- ☐ False

What is the mass of one mole of aluminum?

- ☐ 26.982 g
- ☐ 13 g
- ☐ 53.965 g
- ☐ 14 g

Which is the correct formula for the compound whose percent composition is shown?

Study Guide Mole Test

John William Hill, Dorothy M. Feigl



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