

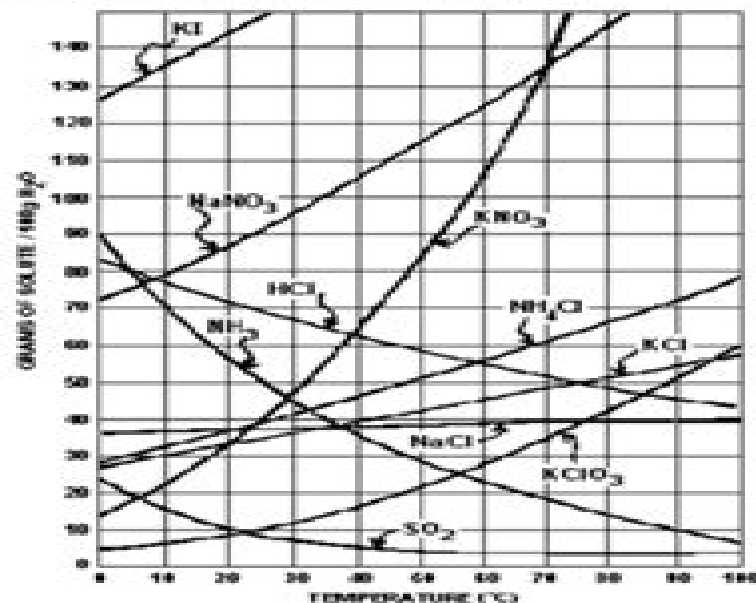
## Solutions Level II

**Section I: Reading Solubility Curves** – Use the solubility curve below to answer the following questions.

1. What relationship exists between solubility and temperature for the ionic substances shown?
2. What relationship exists between solubility and temperature for the only gas,  $\text{SO}_2$ , on the graph?

Indicate if the following solutions are saturated, unsaturated or supersaturated.

3. 20 grams of  $\text{NH}_3$  completely dissolved in 100 grams of water at  $80^\circ\text{C}$ .
4. 80 grams of  $\text{KNO}_3$  completely dissolved in 100 grams of water at  $55^\circ\text{C}$ ?
5. 112 grams of  $\text{NaNO}_3$  completely dissolved in 100 grams water at  $45^\circ\text{C}$ ?
6. 10 grams of  $\text{KClO}_3$  completely dissolved in 100 grams water at  $60^\circ\text{C}$ ?
7. 30g  $\text{KCl}$  in 100g water at  $10^\circ\text{C}$ ?



**Section II: Electrolytes** - Complete the following table. The first one has been done for you as an example.

	Formula?	Ionic, Acid/Base, or Covalent?	Soluble in water?	Electrolyte?
Sodium chloride	NaCl	Ionic	Yes	Yes
1. Acetic Acid				
2. Silver bromide				
3. Hydrofluoric acid				
4. Carbon dioxide				
5. Strontium hydroxide				
6. Magnesium chlorate				
7. Ammonium nitrate				

Section I: 1. Directly related; 2. Inversely related; 3. S; 4. U; 5. S; 6. U; 7. S Section II: 1. Y 2. N 3. Y 4. N 5. Y 6. Y 7. Y

# Study Guide Solubility Graph Answers

**Leo J. Malone**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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