

4. a.
$$\frac{10g \text{ Al} \mid 1 \text{ mol Al} \mid 3 \text{ mol Cu} \mid 63.55g \text{ Cu}}{26.98g \text{ Al} \mid 2 \text{ mol Al} \mid 1 \text{ mol Cu}} = 35.33g \text{ Cu}$$

$$\frac{20g \text{ CuCl}_2 \mid 1 \text{ mol CuCl}_2 \mid 3 \text{ mol Cu} \mid 63.55g \text{ Cu}}{134.45g \text{ CuCl}_2 \mid 3 \text{ mol CuCl}_2 \mid 1 \text{ mol Cu}} = \boxed{9.45g \text{ Cu}}$$

b.
$$\frac{30g \text{ Al} \mid 1 \text{ mol Al} \mid 2 \text{ mol AlCl}_3 \mid 133.33g \text{ AlCl}_3}{26.98g \text{ Al} \mid 2 \text{ mol Al} \mid 1 \text{ mol AlCl}_3} = 148.25g \text{ AlCl}_3$$

$$\frac{100g \text{ CuCl}_2 \mid 1 \text{ mol CuCl}_2 \mid 2 \text{ mol AlCl}_3 \mid 133.33g \text{ AlCl}_3}{134.45g \text{ CuCl}_2 \mid 3 \text{ mol CuCl}_2 \mid 1 \text{ mol AlCl}_3} = \boxed{66.11g \text{ AlCl}_3}$$

c.
$$\frac{5g \text{ Al} \mid 1 \text{ mol Al} \mid 3 \text{ mol Cu} \mid 63.55g \text{ Cu}}{26.98g \text{ Al} \mid 2 \text{ mol Al} \mid 1 \text{ mol Cu}} = 17.67g \text{ Cu}$$

~~10g / 17.67g~~ $10g / 17.67g \times 100 = \boxed{56.6\%}$

d.
$$\frac{10g \text{ CuCl}_2 \mid 1 \text{ mol CuCl}_2 \mid 2 \text{ mol AlCl}_3 \mid 133.33g \text{ AlCl}_3}{134.45g \text{ CuCl}_2 \mid 3 \text{ mol CuCl}_2 \mid 1 \text{ mol AlCl}_3} = 6.61g \text{ AlCl}_3$$

~~5g / 6.61g~~ $5g / 6.61g \times 100 = \boxed{75.6\%}$

- 5.
- a. TRUE
 - b. FALSE
 - c. $\text{AgNO}_3 \Rightarrow 169.91g/\text{mol}$
 - d. 6.02×10^{23} particles = 1 mole of particles

Study Guide Stoichiometry Answers

Paul B. Kelter



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The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

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Student Study Guide to accompany Chemistry Martin Silberberg, Libby Weberg, 2005-01-06

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stressed throughout the text developing chemical thinking and chemical vision and refining problem solving skills Many chapters in this edition has been rewritten and rearranged to vitalize the topics and to include interesting examples analogies and images

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