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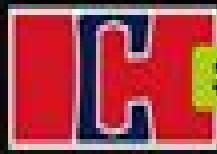
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the mole 4 2 1 cie igcse chemistry revision notes 2022 - Jul 14 2023

web for practical purposes a rounded version of the constant can be used in exams 6.02×10^{23} mol⁻¹ the reason we have mol⁻¹ as the unit is because this is the number of entities per mole of substance for example one mole of sodium na contains 6.02×10^{23} atoms of sodium one mole of hydrogen h₂ contains 6.02×10^{23} molecules of hydrogen