

## Practice Test #21: Nuclear Chemistry

1. Of the following processes, which one does not change the atomic number?

A) alpha emission  ${}_Z^XA \rightarrow {}_{Z-2}^{A-4}B + {}_2^4\alpha$

B) beta emission  ${}_Z^XA \rightarrow {}_{Z+1}^XB + {}_{-1}^0\beta$

C) electron capture  ${}_Z^XA + {}_{-1}^0e(\text{orbital electron}) \rightarrow {}_{Z-1}^XB$

D) gamma emission  ${}_Z^XA^* \rightarrow {}_Z^XA + {}_0^0\gamma$  (only this nuclear reaction does not change the atomic number)

E) positron emission  ${}_Z^XA \rightarrow {}_{Z-1}^XB + {}_{+1}^0e$

2. In balancing the nuclear reaction below, the identity of element "E" is determined from its...



A) atomic weight.

B) mass number.

C) atomic number.

D) number of neutrons.

E) number of electrons.

3. The largest number of stable nuclei have an \_\_\_\_ number of protons and an \_\_\_\_ number of neutrons.

A) even, even

B) odd, odd

C) even, odd

D) odd, even

4. The reaction below is an example of \_\_\_\_\_ decay.



A) alpha

B) beta

C) gamma

D) positron

# Study Guide Nuclear Chemistry Answer Key

**Jianjun Gao**



## Study Guide Nuclear Chemistry Answer Key:

**Chemical Principles Study Guide/Solutions Manual** John Krenos, Joseph Potenza, Carl Hoeger, 2007-01-18 Written for general chemistry courses Chemical Principles helps students develop chemical insight by showing the connection between chemical principles and their applications

**Chemical Principles Student's Study Guide & Solutions Manual** John Krenos, Joseph Potenza, Loretta Jones, Lynn Koplitz, Thomas Spence, 2004-03-19 This combination manual is designed to help students avoid common mistakes and understand the material better The solutions manual section includes detailed answers and explanations to the odd numbered exercises in the text

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**Study Guide/Selected Solutions Manual** Julie R. Frentrop, 2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book

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Student Study Guide/solutions Manual to Accompany Foundations of Inorganic, Organic, and Biological Chemistry Katherine J. Denniston, 1995 This student study guide has been written to accompany Foundations of Inorganic Organic and Biological Chemistry by Caret Denniston and Topping It was designed to complement the text not to be used in place of the text Each chapter of the study guide contains the following sections reorganized and expanded set of learning goals concise chapter summary in chapter solved problems list of key terms review problems Preface

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