

Unit 7 Test Study Guide

(Right Triangles & Trigonometry)

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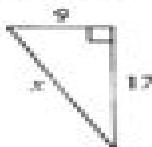
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Topic 1: The Pythagorean Theorem

For any right triangle: $a^2 + b^2 = c^2$, where a and b are legs and c is the hypotenuse.

Directions: Solve for x . Round your answer to the nearest tenth.

1.



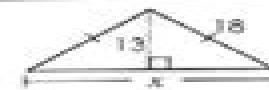
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3.



4.



5. An Olympic-size swimming pool is approximately 50 meters long by 25 meters wide. What distance will a swimmer travel if they swim from one corner to the opposite?

6. A 20-foot ramp is used at the loading dock of a factory. If the base of the ramp is placed 19 feet from the base of the dock, how high is the loading dock?

Topic 2: Pythagorean Theorem Converse & Classifying Triangles

Given a triangle with side lengths a , b , and c :

- If $a^2 + b^2 = c^2$, then it is a **right** triangle.
- If $a^2 + b^2 < c^2$, then it is an **acute** triangle.
- If $a^2 + b^2 > c^2$, then it is an **obtuse** triangle.



Recall: To be a triangle, the sum of the two smaller sides must be greater than the third largest side.

Directions: Classify the triangle with the given side lengths as acute, right, obtuse, or not a triangle.

7. 20, 21, 29

- Not a triangle
- Acute
- Right
- Obtuse

8. 5, 9, 10

- Not a triangle
- Acute
- Right
- Obtuse

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web in order to balance a chemical equation the quantities of each type of element and polyatomic ion that are present in the reactants and the products of the reaction must be determined because polyatomic ions contain multiple atoms and generally react as indivisible units any polyatomic ion that is present in both a reactant and a product in

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web balancing equations the chemical equation described in section 4 1 is balanced meaning that equal numbers of atoms for each element involved in the reaction are represented on the reactant and product sides this is a requirement the equation must satisfy to be consistent with the law of conservation of matter

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web a key step in balancing chemical equations is correctly identifying the formulas of the reactants and products for example consider the reaction between calcium oxide CaO s and H_2O l l2 to form aqueous calcium hydroxide b is it possible to balance the equation if you incorrectly identify the product as CaOH_1aq_2 and if so what is the equation

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web balance a chemical equation when given the unbalanced equation explain the role of the law of conservation of mass in a chemical reaction even though chemical compounds are broken up and new compounds are formed during a chemical reaction atoms in the reactants do not disappear nor do new atoms appear to form the products

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web balance the following chemical equation $\text{Mg} + \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2\text{O}$ note all reactants and products require a coefficient of at least one stuck review related articles videos or use a hint report a problem do 4 problems

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