

Name _____ Date/Period _____ **Key**

Single Replacement Reaction Worksheet

For each reaction use the activity series to complete the reaction. If no reaction will occur, write "no reaction". Make sure you balance your final answer.

1. $\text{Ag} + \text{KNO}_3 \rightarrow$ No reaction
2. $\text{Zn} + 2\text{AgNO}_3 \rightarrow$ $2\text{Ag} + \text{Zn}(\text{NO}_3)_2$
3. $\text{Cl}_2 + 2\text{KI} \rightarrow$ $2\text{KCl} + \text{I}_2$
4. $\text{Cu} + \text{FeSO}_4 \rightarrow$ No reaction
5. $2\text{Fe} + 3\text{Pb}(\text{NO}_3)_2 \rightarrow$ $2\text{Fe}(\text{NO}_3)_3 + 3\text{Pb}$
6. $\text{Cu} + \text{Al}_2(\text{SO}_4)_3 \rightarrow$ No reaction
7. $2\text{Al} + 3\text{Pb}(\text{NO}_3)_2 \rightarrow$ $2\text{Pb} + 2\text{Al}(\text{NO}_3)_3$
8. $\text{Cl}_2 + 2\text{NaI} \rightarrow$ $2\text{NaCl} + \text{I}_2$
9. $\text{Fe} + 3\text{AgC}_2\text{H}_3\text{O}_2 \rightarrow$ $3\text{Ag} + \text{Fe}(\text{C}_2\text{H}_3\text{O})_2$
10. $2\text{Al} + 3\text{CuCl}_2 \rightarrow$ $3\text{Cu} + 2\text{AlCl}_3$
11. $\text{Br}_2 + \text{CaI}_2 \rightarrow$ $\text{CaBr}_2 + \text{I}_2$
12. $2\text{Fe} + 3\text{CuSO}_4 \rightarrow$ $3\text{Cu} + \text{Fe}_2(\text{SO}_4)_3$
13. $\text{Cl}_2 + \text{MgI}_2 \rightarrow$ $\text{MgCl}_2 + \text{I}_2$

Most

Most

Least

Least

Activity Series	
Metals	Nonmetals
Li	F_2
Rb	Cl_2
K	Br_2
Cs	I_2
Ba	
Sr	
Ca	
Na	
Mg	
Al	
Ti	
Mn	
Zn	
Cr	
Fe	
Co	
Ni	
Sn	
Pb	
H_2	
Cu	
Ag	
Au	

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