

CHAPTER **21** STUDY GUIDE

# Hydrocarbons

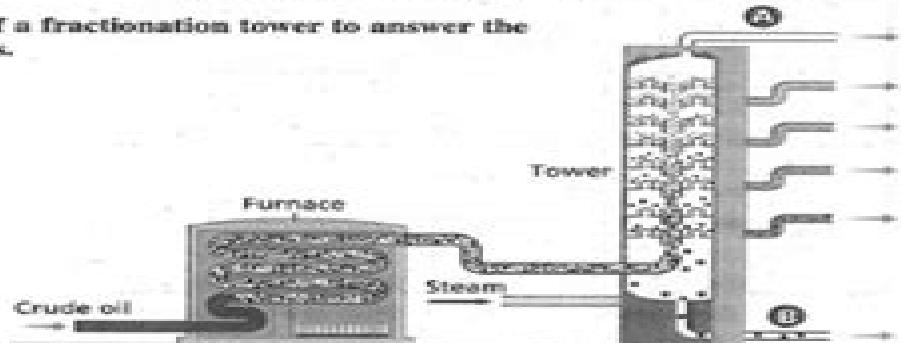
## Section 21.1 Introduction to Hydrocarbons

In your textbook, read about natural sources of hydrocarbons and rating gasoline.

In the space at the left, write the word or phrase in parentheses that correctly completes the statement.

1. (Petroleum, Natural gas) is a mixture of thousands of organic compounds.
2. The boiling of petroleum and collection of its components is called (sedimentation, fractional distillation).
3. In the process known as (cracking, knocking), heavier petroleum fractions are converted to gasoline by breaking their large molecules into smaller ones.
4. A gasoline's ability to burn evenly and prevent knocking is expressed by its (hexane, octane) rating.

Use the diagram of a fractionation tower to answer the following questions.



5. How does the temperature inside the tower vary from bottom to top? \_\_\_\_\_
6. In what physical state is the material collected from pipe A? \_\_\_\_\_
7. What is the material collected from pipe A used for? \_\_\_\_\_
8. What is the material collected from pipe B used for? \_\_\_\_\_
9. Which pipe, A or B, collects hydrocarbons with higher boiling points? \_\_\_\_\_
10. Which pipe, A or B, collects smaller hydrocarbons? \_\_\_\_\_

# Study Guide Hydrocarbons Answer Keys

**Arshad Iqbal**

## **Study Guide Hydrocarbons Answer Keys:**

Organic Chemistry Study Guide with Solutions Manual Neil E. Schore,2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer

**Study Guide and Solutions Manual** Neil E. Schore,2002-08-02

*Study Guide with Selected Solutions* Karen Timberlake,2002-08-06

Written by the author the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions to odd numbered problems

*Study Guide for Chemical Principles* Thomas Elliott Taylor,1979

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Katherine J. Denniston,Denniston Katherine,2006

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