

# 10.1 Periodic Trends



## PERIODIC TRENDS: FROM CONCEPT TO APPLICATION

### Section 10.1 continued

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

_____
_____
_____
_____
_____

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

#### Column A

- \_\_\_\_\_ 16. It varies on the periodic table.
- \_\_\_\_\_ 17. It does not lie periodic table.
- \_\_\_\_\_ 18. Group 1 elements.
- \_\_\_\_\_ 19. Elements that are strong oxidizing agents.
- \_\_\_\_\_ 20. Group 8 elements.

#### Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write one of the statements in terms of the statement in italics, change the underlined word or phrase to make it true.

- \_\_\_\_\_ 21. There are two representative elements of elements.
- \_\_\_\_\_ 22. Most of the elements of the periodic table are metals.
- \_\_\_\_\_ 23. Group 1A elements except for hydrogen are metals on the left.
- \_\_\_\_\_ 24. Group 2A elements are the alkaline earth metals.
- \_\_\_\_\_ 25. Group 7A elements are highly reactive nonmetals known as halogens.
- \_\_\_\_\_ 26. Group 8A elements are very reactive elements known as noble gases.
- \_\_\_\_\_ 27. Alkaloids have properties of both metals and nonmetals.

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