

<b>Activity A:</b> <b>Phase changes</b>	<b>Get the Gizmo ready:</b> <ul style="list-style-type: none"> <li>Click <b>Reset</b> (  ) and select <b>Macro view</b>.</li> <li>Set the <b>Water temperature</b> to <b>10 °C</b>.</li> <li>Set the <b>Ice volume</b> to <b>0 cm<sup>3</sup></b>.</li> </ul>	
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**Question: How is temperature related to phase changes?**

**1. Predict:** Based on your prior knowledge, predict the following:

- At what temperature will water change from a liquid to a solid (**freeze**)?
- At what temperature will water change from a solid to a liquid (**melt**)?
- At what temperature will water change from a liquid to a gas (**boil**)?

0

0

100

**2. Investigate:** Use the Gizmo to explore phase changes. Use the **Add/remove heat energy** slider to control the water temperature. Record your observations in your notes, then answer the questions below:

- At what temperature does water freeze?
- At what temperature does ice melt?
- At what temperature does water boil?

0

0

10

This is the **freezing point**.

This is the **melting point**.

This is the **boiling point**.

**3. Observe:** Set up the Gizmo to observe freezing. What do you notice about the temperature while the water is in the process of freezing?

Remains unchanged

**4. Explore:** Use the Gizmo to investigate melting and boiling. Does the temperature change while either of these phase changes is occurring?

No it doesn't

**5. Interpret:** Select the **GRAPH** tab to see a graph of temperature vs. time. Click the **"-"** button until the whole graph is visible. What does the graph look like during a phase change?

A horizontal line

**6. Extend your thinking:** Why do you think the temperature does not change much during a phase change? If possible, discuss your answer with your classmates and teacher.

Utilizing the energy to alter the nature of matter

# Study Guide Phase Changes Answer Key

**R Barnett**

## **Study Guide Phase Changes Answer Key:**

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