

Name _____ Period _____ Date _____

STAAR Science Test Review Worksheet

Part 1: Matter & Energy

1. Describe the properties (charge, mass, location) of the three sub-atomic particles. _____

2. Using a periodic table, how can you determine the number of protons (or electrons) in an atom of an element? _____

3. Using a periodic table, how can you determine how many neutrons are found in the most common isotope of an atom for a particular element? _____

4. Using a periodic table, how can you determine the number of electron orbitals or shells in an atom of a particular element? _____

5. What is the maximum number of electrons that can be in each of the first four orbitals of an atom? _____
How many electrons can be in the outer most orbital? _____
6. Using a periodic table, how can you determine the number of valence electrons in an atom of a particular element? _____

7. Which elements are not reactive? _____
Which elements are most reactive? _____
What determines the reactivity of elements? _____

8. How can you tell whether a chemical equation is balanced? _____

Staar Practice Science Answer Guide

Responsive Education Solutions

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the 'Responsive Education Solutions' text.

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