

1. Light and Quantum Energy: The Wave Nature of Light

The Photoelectric Effect (Einstein's Model)

- 1. Did not contradict the wave's characteristics arranged under special circumstances
- 2. Did not address the question of why the negative's electron ejection was not related to the wave's periodicity (frequency)
- 3. Did not conclude that the distribution and distribution of characteristics among the various elements

Wave Properties

- 1. Wave properties of light are characterized by:
 - 1. Frequency: the number of waves that pass a point per second
 - 2. Wavelength: the distance between two consecutive crests or troughs
 - 3. Amplitude: the height of the wave from the equilibrium position
 - 4. Phase: the position of a wave relative to a reference point
- 2. Wave properties are related to the speed of light:
 - 1. $c = \lambda \cdot f$ (where c is the speed of light, λ is the wavelength, and f is the frequency)



Wave Properties

- 1. All waves travel at the same speed ($c = 3.0 \times 10^8$ m/s)
- 2. The frequency of a wave is related to the wavelength by:
 - 1. $f = \frac{c}{\lambda}$ (where f is the frequency, c is the speed of light, and λ is the wavelength)
- 3. The wave energy is related to the frequency:
 - 1. $E = hf$ (where E is the energy, h is Planck's constant, and f is the frequency)
- 4. The wave energy is related to the wavelength:
 - 1. $E = \frac{hc}{\lambda}$ (where E is the energy, h is Planck's constant, and λ is the wavelength)

Wave Properties: Wavelength



Study Guide Light And Quantized Energy

Chih Tang Sah



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Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement

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