

# States of Matter: Phase Changes



Phase	Description	Illustration
Evaporation	<b>Liquid</b> → _____ <ul style="list-style-type: none"> <li>Water changes into a gas called _____</li> <li>Heat energy is added</li> <li>Particles move faster and faster</li> <li>Examples include a _____ disappearing and clothes drying</li> </ul>	
Condensation	<b>Gas</b> → _____ <ul style="list-style-type: none"> <li>Water vapor turns into droplets of liquid _____</li> <li>Heat energy is removed</li> <li>Particles _____</li> <li>Examples include morning dew and a foggy mirror after a hot shower</li> </ul>	
Freezing	<b>Liquid</b> → _____ <ul style="list-style-type: none"> <li>Heat energy is _____</li> <li>(Liquid) water turns into ice</li> <li>Particles slow down</li> <li>Examples include rain turning into _____ and juice freezing into a popsicle</li> </ul>	
Melting	<b>Solid</b> → _____ <ul style="list-style-type: none"> <li>Heat energy is _____</li> <li>Ice turns into (liquid) water</li> <li>Particles move _____</li> <li>Examples include a snowman thawing out and wax dripping down a candle</li> </ul>	

- Matter can change from one phase to another with the addition or removal of \_\_\_\_\_
- Matter changing from one state to another is a \_\_\_\_\_

# Study Guide Phase Changes

**Joseph J. Boyle, Douglas C. Giancoli**

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