

Matter—Properties and Changes

Lesson 2.1 Properties of Matter

In your textbook, read about physical properties and chemical properties of matter.

Use each of the terms below just once to complete the passage.

chemical density	mass properties	physical substance
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Matter is anything with (1) MASS and volume. A (2) SUBSTANCE is a form of matter with a uniform and unchanging composition. Substances have specific, unchanging (3) PROPERTIES that can be observed. Substances have both physical and chemical properties. (4) PHYSICAL properties can be observed without changing a substance's chemical composition. Color, hardness, and (5) DENSITY are examples. Other properties cannot be observed without changing the composition of a substance. These are called (6) CHEMICAL properties. An example is the tendency of iron to form rust when exposed to air.

Label each property as either *physical* or *chemical*.

CHEM	7. Chemical formula H ₂ O
CHEM	8. Forms green carbonate when exposed to moist air
CHEM	9. Remains unchanged when in the presence of nitrogen
PHYS	10. Colorless
PHYS	11. Solid at normal temperatures and pressures
CHEM	12. Ability to combine with another substance
PHYS	13. Melting point
PHYS	14. Liquid at normal temperatures and pressures
PHYS	15. Boiling point is 100°C
PHYS	16. Conducts electricity

Study Guide Properties Of Matter Answers

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Study Guide Properties Of Matter Answers:

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03

The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step by step examples. Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning. This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

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- Chapter 5: Conclusion

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4. In chapter 3, the author will examine the practical applications of Study Guide Properties Of Matter Answers in daily life. This chapter will showcase real-world examples of how Study Guide Properties Of Matter Answers can be effectively utilized in everyday scenarios.
5. In chapter 4, this book will scrutinize the relevance of Study Guide Properties Of Matter Answers in specific contexts. This chapter will explore how Study Guide Properties Of Matter Answers is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about Study Guide Properties Of Matter Answers. This chapter will summarize the key points that have been discussed throughout the book.

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