



Chemistry

Paper Review and Study Guide

Periodic Table and Law

4-page Review +

39 Question Study Guide

Periodic Table and Law

Development of the Modern Periodic Table

Early Periodic Tables:

- Dmitri Mendeleev** (1869): Dmitri Mendeleev published the first periodic table. He arranged elements by increasing atomic mass and noticed that certain properties periodically repeated. Mendeleev's table left gaps for elements that had not yet been discovered, predicting their properties accurately.
- Luther Meyer** (independently): Luther Meyer published a similar periodic table around the same time as Mendeleev.

Modern Periodic Table:

- Moseley's Principle** (1913): Henry Moseley discovered the atomic number of each element, leading to the modern definition of the periodic law. The modern periodic table is arranged by increasing atomic number rather than atomic mass.

The periodic table is a powerful tool for understanding the behavior of elements. It's organized by increasing atomic number, with elements arranged in rows called periods and columns called groups. Elements within the same group share similar properties, while elements in the same period share the same number of electron shells. The elements are further divided into metals, nonmetals, and metalloids.

Periodic Law:

The periodic law states that the properties of the elements are a function of their atomic number. This means that when elements are arranged by increasing atomic number, elements with similar properties appear at regular intervals.

Periods and Groups:

Periods: Horizontal rows of elements. There are 7 periods in the modern periodic table. Elements in the same period have the same number of electron shells.

Groups: Vertical columns on the periodic table. Elements in the same group have similar chemical properties.

Periodic Trends:

Atomic size, ionization energy, and electronegativity are key periodic trends. Atomic size increases from top-left to bottom-right. Ionization energy and electronegativity increase from bottom-left to top-right.

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Study Guide Review The Periodic Law

Robert K. Wismer



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web 1 1 given two vectors a and b do the equations $v \cdot a = b$ and $v \cdot a = k$ determine the vector v uniquely if so find an explicit formula of v in terms of a and b answer the answer is yes clearly if a and b are not orthogonal then there is no solution so assume a and b are orthogonal vectors

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web chapter 1 vector analysis 1 1 definitions elementary approach in science and engineering we frequently encounter quantities that have magnitude and magnitude only mass time and temperature these we label scalar quantities which remain the same no matter what coordinates we use

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web nov 16 2022 basic concepts in this section we will introduce some common notation for vectors as well as some of the basic concepts about vectors such as the magnitude of a vector and unit vectors we also illustrate how to

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web opening chapters define vector addition and subtraction show how to resolve and determine the direction of two or more
vectors and explain systems of coordinates vector equations of a plane and straight line relative velocity and

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web components of vectors for the analytical method of vector addition and subtraction we use some simple geometry and
trigonometry instead of using a ruler and protractor as we did for graphical methods however the graphical method will still
come in handy to visualize the problem by drawing vectors using the head to tail method

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web numerically the solution is $\vec{d} = \vec{d}_0 + \vec{d}_1 + \vec{d}_2 + \vec{d}_3 + \vec{d}_4 + \vec{d}_5 + \vec{d}_6$ find the unit vectors that point from a to the other points b
to g in question 2 express each vector in component $\hat{i}\hat{j}$ notation

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web 2 review of vector analysis electromagnetic field theory is the study of forces between charged particles resulting in
energy conversion or signal transmission and reception these forces vary in magnitude and direction with time and
throughout space so that the theory is a heavy user of vector differential and integral calculus