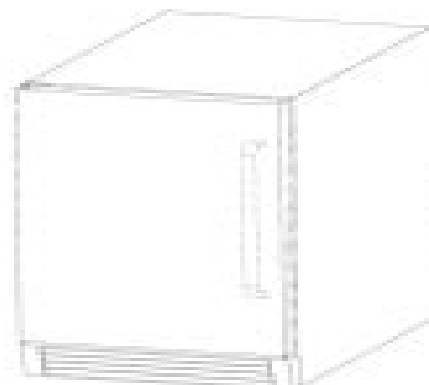


SUMMIT

C O M M E R C I A L

Automatic Undercounter Ice Maker *Máquina automática para hacer hielo* Machine à glaçon automatique



User's Manual (pages 1 - 29)
Manual del usuario (páginas 30 a 57)
Mode d'emploi (pages 58 à 86)

Please see ice maker is standing upright 24 hours prior to plug-in.

Antes de que la máquina esté parada en posición vertical 24 horas antes que sea conectada.

Assurez-vous que la machine à glaçons est debout 24 heures avant d'être branchée.

SUMMIT® APPLIANCE DIVISION Model/Modelo/Modèle

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Bronx, NY 10474

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Brain Guide



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Do-it-yourself Repair Manual Whirlpool Corporation,1996 *Kismile Nugget Ice Maker User Guide* Rose J Cravens,2025-08-13 Over 10 000 new Kismile owners can't be wrong In just three days they discovered the secret to mastering their machine Are you still guessing Don't be left behind with a silent underperforming machine and a growing fear of it breaking down The official manual is just a piece of paper it won't show you how to truly unlock your ice maker's potential This book is the key that thousands of savvy owners have already found It's not just about what's in the box it's about the transformation from a nervous newbie to a confident pro With this guide you'll join a community that has already left behind The frustration of thin watery ice for an endless supply of perfectly crunchy nuggets The anxiety of a broken machine for the confidence of a pro who can troubleshoot any problem in minutes The chore of manual cleaning for a simple automated maintenance plan that protects

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for the nitration 4.01 ml of nitric acid 4.00 ml of sulfuric acid and 4.56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4.56 ml of bromobenzene was used the amount of product predicted to be recovered was 8.7533 g

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1-bromo-4-nitrobenzene

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture HNO_3 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accommodate their electrons demonstration of creating the p arenium cation by nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion NO_2^+ which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4 0ml of nitric acid and 4 0ml of sulfuric acid and will then react with 4 5ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

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