

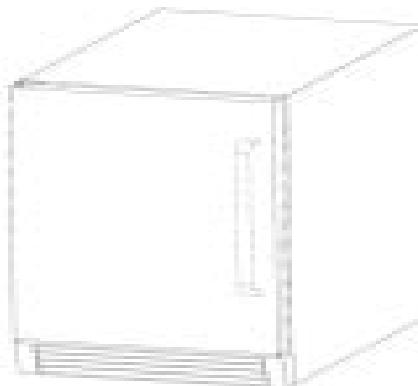
SUMMIT

C O M M E R C I A L

Automatic Undercounter Ice Maker

Máquina automática para hacer hielo

Machine à glaçon automatique



User's Manual (pages 1 - 29)
Manual del usuario (páginas 30 a 57)
Mode d'emploi (pages 58 à 86)

Be sure ice maker is standing upright 24 hours prior to plug-in.

Asegúrese que la máquina esté parada en posición vertical 24 horas antes que sea conectada.

Assurez-vous que la machine à glaçons est debout 24 heures avant d'être branchée.

SUMMIT APPLIANCE DIVISION Model/Modelo/Modèle

Polka Search, Inc.

An ISO 9001:2008 registered company

770 Garrison Avenue

Bronx, NY 10474

Website: www.summitappliance.com

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Summit Bim44 Ice Maker Manual

Brain Guide

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Table of Contents Summit Bim44 Ice Maker Manual

1. Understanding the eBook Summit Bim44 Ice Maker Manual
 - The Rise of Digital Reading Summit Bim44 Ice Maker Manual
 - Advantages of eBooks Over Traditional Books
2. Identifying Summit Bim44 Ice Maker Manual
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Summit Bim44 Ice Maker Manual
 - User-Friendly Interface
4. Exploring eBook Recommendations from Summit Bim44 Ice Maker Manual
 - Personalized Recommendations
 - Summit Bim44 Ice Maker Manual User Reviews and Ratings
 - Summit Bim44 Ice Maker Manual and Bestseller Lists
5. Accessing Summit Bim44 Ice Maker Manual Free and Paid eBooks
 - Summit Bim44 Ice Maker Manual Public Domain eBooks
 - Summit Bim44 Ice Maker Manual eBook Subscription Services
 - Summit Bim44 Ice Maker Manual Budget-Friendly Options
6. Navigating Summit Bim44 Ice Maker Manual eBook Formats

- ePub, PDF, MOBI, and More
- Summit Bim44 Ice Maker Manual Compatibility with Devices
- Summit Bim44 Ice Maker Manual Enhanced eBook Features

7. Enhancing Your Reading Experience

- Adjustable Fonts and Text Sizes of Summit Bim44 Ice Maker Manual
- Highlighting and Note-Taking Summit Bim44 Ice Maker Manual
- Interactive Elements Summit Bim44 Ice Maker Manual

8. Staying Engaged with Summit Bim44 Ice Maker Manual

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Summit Bim44 Ice Maker Manual

9. Balancing eBooks and Physical Books Summit Bim44 Ice Maker Manual

- Benefits of a Digital Library
- Creating a Diverse Reading Collection Summit Bim44 Ice Maker Manual

10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

11. Cultivating a Reading Routine Summit Bim44 Ice Maker Manual

- Setting Reading Goals Summit Bim44 Ice Maker Manual
- Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Summit Bim44 Ice Maker Manual

- Fact-Checking eBook Content of Summit Bim44 Ice Maker Manual
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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then actual is your actual findings yield actual theoretical x 100 what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 °

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1 bromo 4 nitrobenzene

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study with quizlet and memorize flashcards containing terms like reacting sulfuric and nitric acid together produces the nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accomodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion no2 which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4 0ml of nitric acid and 4 0ml of sulfuric acid and will then react with 4 5ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

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