

# Online ACT

## Science

 40 Questions  40 Minutes

### Interpretation of Data (36%-47%)

- Scientific Data Analysis
- Trends in Data
- Scientific Graphs
- Mathematical Reasoning

### Scientific Investigation (17%-31%)

- Experimental Tools
- Experimental Procedures
- Experiment Modifications
- Experiment Comparisons
- Experiment Extensions

### Evaluation of Models, Inferences, and Experimental Results (22%-36%)

- Validity of Scientific Information
- Formulation of Predictions
- Formulation of Conclusions

# Paper-and-Pencil ACT

## Science

 40 Questions  35 Minutes

### Interpretation of Data (40%-50%)

- Scientific Data Analysis
- Trends in Data
- Scientific Graphs
- Mathematical Reasoning

### Scientific Investigation (20%-30%)

- Experimental Tools
- Experimental Procedures
- Experiment Modifications
- Experiment Comparisons
- Experiment Extensions

### Evaluation of Models, Inferences, and Experimental Results (25%-35%)

- Validity of Scientific Information
- Formulation of Predictions
- Formulation of Conclusions

# Act Practice On Sale Tutorial

**Yijin Wang**



## Act Practice On Sale Tutorial:

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isobars differ in atomic number or number of protons but have the same mass number an example of a series

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web isotope isobar isotopes are atomic structures of same elements having a different mass number atomic mass isobars are different chemical elements having same atomic mass atomic numbers of all isotopic forms of a single element are equal

atomic numbers of isobars vary from each other they are the same chemical element but their forms are

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web solution verified by toppr isotopes same atomic number but different mass number example  $^{12}_6\text{C}$   $^{14}_6\text{C}$  isobars isobars are atoms of different elements having same mass number these have equal number of nucleons but different number of protons neutrons and electrons was this answer helpful 0 0 similar questions define isobars and