

The Working Cell

Lecture 5: Cellular Metabolism

metabolism → culmination of all the "work" that your cell does

- the use of energy
- importance of enzymes
- building or breaking down molecules
- movement of fluids or molecules

energy → capacity to do work / cause change

2 TYPES OF ENERGY

- Potential energy — stored energy
- Kinetic energy — energy in motion

conservation of Energy

Law of thermodynamics

- energy cannot be destroyed nor created

heat → energy that is lost

- caused by the friction between atoms & molecules

Chemical Energy — where energy comes from

- molecule in food can be broken down & digested to provide energy & nutrients

laws of thermodynamics

ENTROPY — random nature of energy

- measures "randomness" (disorder) of energy

- Heat is produced in most every reaction
 - Exothermic reactions → releases energy (exergonic)
 - Endothermic reactions → absorbs energy (endergonic)
- All transfers of energy loses heat
 - Energy transfer loses **90%** of energy as heat

enzymes

→ made of proteins
→ proteins that perform cell reaction

↳ shape determines functions

- Substrate — where the specific molecule that an enzyme acts on



active site — fit the substrate

PROPERTIES / FEATURES

- made of proteins
- speed up reaction
- don't change reaction
- unchanged
- reusable

ENZYME INHIBITION

- slows down from doing its job



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Jacquelyn L. Banasik

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