

# The Working Cell

## Lecture 5: Cellular Metabolism

**metabolism** → culmination of all the "work" that your cell does

- the use of energy
- importance of enzymes
- building or breaking down molecules
- movement of fluids or molecules

**energy** → capacity to do work / cause change

### 2 TYPES OF ENERGY

- \* **Potential energy** — stored energy
- \* **Kinetic energy** — energy in motion

### Conservation of Energy

#### Law of thermodynamics

→ energy cannot be destroyed nor created

**Heat** → energy that is lost  
→ caused by the friction between atoms & molecules

**Chemical Energy** — where energy comes from  
— molecule in food can be broken down & digested to provide energy & nutrients

### Laws of thermodynamics

**ENTROPY** — random nature of energy  
— measures "randomness" (disorder) of energy

- Heat is produced in most every reaction
  - \* **Exothermic reactions** → releases energy (exergonic)
  - \* **Endothermic reactions** → absorbs energy (endergonic)
- All transfers of energy loses heat
  - \* Energy transfer loses 90% of energy as heat

**Enzymes** → made of proteins  
→ proteins that perform cell reaction

↳ shape determines functions

- \* **Substrate** — where the specific molecule that an enzyme acts on



active site — fit the substrate

### PROPERTIES / FEATURES

- made of proteins
- speed up reaction
- don't change reaction
- unharmed
- reusable

### ENZYME INHIBITION

— slow down from doing its job



# Working Cell Study Guide Answers

**Jacquelyn L. Banasik**



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