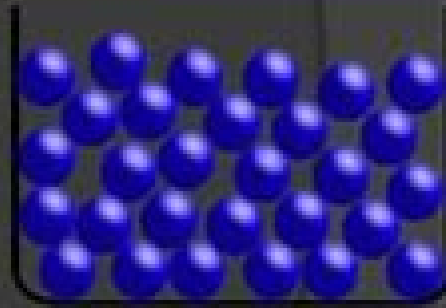


Second Phase of Matter

Liquids

- Liquids are not as tightly packed as solids and more tightly packed than gases.
- Liquids have more energy than solids, so the atoms move around more.
- Liquids have particles that slide over and around each other. (flow)
- Liquids will take the shape of whatever container they are put in.



The Molecular Theory Of Gases And Liquids

Richard D. Kleeman



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a point in the substance at which the forces of the surrounding molecules neutralize each other The total average velocity corresponding to the whole path of a molecule is usually much greater than the foregoing velocity in a liquid and dense gas on account of the effect of the molecular forces of attraction and repulsion About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books Find more at www.forgottenbooks.com This book is a reproduction of an important historical work Forgotten Books uses state of the art technology to digitally reconstruct the work preserving the original format whilst repairing imperfections present in the aged copy In rare cases an imperfection in the original such as a blemish or missing page may be replicated in our edition We do however repair the vast majority of imperfections successfully any imperfections that remain are intentionally left to preserve the state of such historical works

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