

TEACHING TRANSPARENCY 22

Microscopic Chapter 2
Section 1.1

Formation of ions

- Q 1. What are cations in the following context?
Electron deficient
- Q 2. How do cations appear in the last table of the last experiment (anion)? Chlorine (anion)
positive electron deficient cation
- Q 3. What is the reason for the loss of electron in cations? Electron deficiency, positive (anion) (anion)
Electron deficiency
- Q 4. What kind of change does the formation of anion cause a shift in the electron? What is the reason for this?
Electron deficient cation
- Q 5. What kind of change does the formation of anion cause a shift in the electron? What is the reason for this?
Electron deficient cation
- Q 6. In the above chemical composition of binary compounds, the number of atoms of each element is the same.
Electron deficient cation
- Q 7. In the above chemical composition of binary compounds, the number of atoms of each element is the same.
Electron deficient cation
- Q 8. In the above chemical composition of binary compounds, the number of atoms of each element is the same.
Electron deficient cation
- Q 9. In the above chemical composition of binary compounds, the number of atoms of each element is the same.
Electron deficient cation
- Q 10. What is the chemical composition of the following binary compounds? How are they classified?
Electron deficient cation
- Q 11. What is the chemical composition of the following binary compounds? How are they classified?
Electron deficient cation

Teaching Transparency 22

**Timothy Rasinski, Nancy Padak, Rick
M. Newton, and Evangeline Newton**

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Houghton Mifflin Teacher's Book Houghton Mifflin Company, 1991

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