

TEACHING TRANSPARENCY 22

Background Information
Background

Formation of ions

- When an atom loses or gains electrons it becomes an ion
- An ion has a different net value of the net negative charges from the net positive charges
- When the atom has lost an electron it is called a cation because it has a positive net charge
- When the atom has gained an electron it is called an anion because it has a negative net charge
- Positive charges always have a higher value than negative charges because the positive charge is stronger
- In the diagram above the electrons are now more attracted to the positive charge and repel each other
- In the electron configuration of Sodium when they remove a negative charge there are positive charges
- In the same electron configuration of Sodium when it gains the electrons are still there but they are now negative charges
- In the same electron configuration of Sodium when the Sodium is given a positive charge there are positive charges
- In the same electron configuration of Sodium when the Sodium is given a negative charge there are negative charges
- When in the electron configuration of Sodium when the Sodium is given a positive charge there are positive charges
- When in the electron configuration of Sodium when the Sodium is given a negative charge there are negative charges
- When in the electron configuration of Sodium when the Sodium is given a positive charge there are positive charges
- When in the electron configuration of Sodium when the Sodium is given a negative charge there are negative charges

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**Timothy Rasinski, Nancy Padak, Rick
M. Newton, and Evangeline Newton**

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