

Activity A: Phase changes	Get the Gizmo ready: <ul style="list-style-type: none"> Click Reset () and select Macro view. Set the Water temperature to 10 °C. Set the Ice volume to 0 cm³. 	
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Question: How is temperature related to phase changes?

1. Predict: Based on your prior knowledge, predict the following:

- At what temperature will water change from a liquid to a solid (**freeze**)?
- At what temperature will water change from a solid to a liquid (**melt**)?
- At what temperature will water change from a liquid to a gas (**boil**)?

0

0

100

2. Investigate: Use the Gizmo to explore phase changes. Use the **Add/remove heat energy** slider to control the water temperature. Record your observations in your notes, then answer the questions below:

- At what temperature does water freeze?
- At what temperature does ice melt?
- At what temperature does water boil?

0

0

10

This is the **freezing point**.

This is the **melting point**.

This is the **boiling point**.

3. Observe: Set up the Gizmo to observe freezing. What do you notice about the temperature while the water is in the process of freezing?

Remains unchanged

4. Explore: Use the Gizmo to investigate melting and boiling. Does the temperature change while either of these phase changes is occurring?

No it doesn't

5. Interpret: Select the **GRAPH** tab to see a graph of temperature vs. time. Click the **"-"** button until the whole graph is visible. What does the graph look like during a phase change?

A horizontal line

6. Extend your thinking: Why do you think the temperature does not change much during a phase change? If possible, discuss your answer with your classmates and teacher.

Utilizing the energy to alter the nature of matter

Study Guide Phase Changes Answer Key

Wenbin Ji

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