

Chemistry Spring Final Study Guide

Unit 11: Acids & Bases

Try to find the correct substance and determine the properties of Acids & Bases.

Acids	Bases
1. Hydrochloric acid	1. Sodium hydroxide
2. Sulfuric acid	2. Potassium hydroxide
3. Nitric acid	3. Ammonium hydroxide
4. Acetic acid	4. Calcium hydroxide
5. Phosphoric acid	5. Magnesium hydroxide
6. Hydrofluoric acid	6. Barium hydroxide
7. Perchloric acid	7. Strontium hydroxide
8. Chloric acid	8. Rubidium hydroxide
9. Bromic acid	9. Cesium hydroxide
10. Iodic acid	10. Francium hydroxide

Unit 12: Solubility

Draw a line on the graph and circle the substance for solubility between Temperature, Volume, and Pressure below.

1. **Gas Solubility:** Round your answers to 2 places after the decimal. Write how to change your units. There are 10 mm in 1 cm.
 a) How many grams of gas will dissolve in 100 mL of water at 20°C? b) 100 mL of gas at 20°C.

2. **Gas Solubility:** Round your answers to 2 places after the decimal. Write how to change your units. There are 10 mm in 1 cm.
 a) How many grams of gas will dissolve in 100 mL of water at 20°C? b) 100 mL of gas at 20°C.

Unit 13: Stoichiometry

Balance the following:

1. $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$

2. $\text{H}_2 + \text{Cl}_2 \rightarrow \text{HCl}$

3. $\text{H}_2 + \text{Br}_2 \rightarrow \text{HBr}$

4. $\text{H}_2 + \text{I}_2 \rightarrow \text{HI}$

5. $\text{H}_2 + \text{F}_2 \rightarrow \text{HF}$

6. $\text{H}_2 + \text{S}_8 \rightarrow \text{H}_2\text{S}$

7. $\text{H}_2 + \text{P}_4 \rightarrow \text{PH}_3$

8. $\text{H}_2 + \text{As}_4 \rightarrow \text{AsH}_3$

9. $\text{H}_2 + \text{Sb}_2\text{S}_3 \rightarrow \text{H}_2\text{S}$

10. $\text{H}_2 + \text{Bi}_2\text{S}_3 \rightarrow \text{H}_2\text{S}$

Chemistry Spring Final Study Guide

Name: _____

Unit 1: Measurement Analysis

Mark the best number from those provided to each scenario.

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Sasha Hu



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