

# STUDY GUIDE

Microsoft Office  
Specialist: Associate  
(Microsoft 365 Apps)

## MO 110

# Study Guide Microsoft

**James Pyles**



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web assessment chapter test b chapter acids and bases part i in the space provided write the letter of the term or phrase that best completes each statement or best answers each question 1 which of the following is an oxyacid a hcl b h<sub>2</sub>o c h<sub>2</sub>s d h<sub>2</sub>so<sub>4</sub>  
2 which acid is produced in the stomach a hno<sub>3</sub> b ch<sub>3</sub>cooh c

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web apr 20 2016 a stronger acid c weaker acid b stronger base d weaker base 6 proton transfer reactions favor production of the a stronger acid and stronger base b weaker acid and weaker base c stronger acid and weaker base d weaker acid and stronger base 7 aqueous solutions of most bases contain a hydroxide ions and

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web correct answer d ccl 3 cooh explanation strong acid means weak conjugate base ccl 3 coo chcl 2 coo ch<sub>2</sub>clcoo ch 3 coo therefore the strongest acid is ccl 3 cooh q2 lime water is a cao b ca oh 2 c caco 3 d caci 2 correct answer b ca oh 2 explanation

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web jul 22 2023 correct answer c  $\text{KH}_2\text{PO}_4$  explanation  $\text{KH}_2\text{PO}_4$  is an acid salt because it is derived from a weak acid phosphoric acid and a strong base potassium hydroxide it contains the hydrogen ion from the acid making it acidic but also has the ability to form basic ions due to the presence of the potassium ion

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web the ionization of weak acids and weak bases many acids and bases are weak that is they do not ionize fully in aqueous solution a solution of a weak acid in water is a mixture of the nonionized acid hydronium ion and the conjugate base of the acid with the nonionized acid present in the greatest concentration

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web oct 10 2022 correct answer d salt water explanation neutralization is a chemical reaction between an acid and a base resulting in the formation of a salt and water the product of neutralization is therefore salt water as the acid and base combine to form a salt compound and water molecules rate this question 4

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