

CHAPTER 9 STUDY GUIDE

Chemical Reactions

Section 9.1 Reactions and Equations

In your textbook, read about evidence of chemical reactions.

For each statement, write *yes* if evidence of a chemical reaction is present. Write *no* if there is no evidence of a chemical reaction.

- _____ 1. A tomato smells rotten.
- _____ 2. A drinking glass breaks into smaller pieces.
- _____ 3. A piece of ice melts.
- _____ 4. Drain cleaner is mixed with water and the solution becomes warm.
- _____ 5. Candle wax burns.
- _____ 6. Molten candle wax solidifies.
- _____ 7. Green leaves turn yellow and red as the seasons change.
- _____ 8. Baking powder produces a gas that makes a cake rise.

In your textbook, read about how to represent chemical reactions and how to balance chemical equations.

Use the terms below to complete the passage. Each term may be used once, more than once, or not at all.

arrow	plus sign	(s)	(l)
reactant	product	(g)	(aq)

The fuel for the space shuttle is hydrogen, which burns in oxygen to produce water vapor and energy. In this chemical reaction, hydrogen is a(n) **(9)** _____, oxygen is a(n) **(10)** _____, and water vapor is a(n) **(11)** _____. In a chemical equation for this reaction, a(n) **(12)** _____ is used to separate hydrogen and oxygen from water vapor and energy. A(n) **(13)** _____ is used to separate the symbols for hydrogen and oxygen. A(n) **(14)** _____ symbol is used to tell the state of hydrogen in the reaction, a(n) **(15)** _____ symbol is used for the state of oxygen, and a(n) **(16)** _____ symbol is used for the state of water vapor.

Study Guide Section 2 Chemical Reactions

Wolfgang Guggemos



Study Guide Section 2 Chemical Reactions:

Robinson Chemistry Study Guide Robinson,1992 **Organic Chemistry Study Guide with Solutions Manual** Neil E. Schore,2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer **Study Guide and Solutions Manual** Neil E. Schore,2002-08-02 *Study Guide for Chemical Principles* Thomas Elliott Taylor,1979 Resources in Education ,1984 **Chemistry of Matter** ,1993 As the reader reads this textbook he will learn about the interactions of matter that can occur in a test tube in nature and even inside himself P 9 **Chemistry** Oxtoby,1994 **Holtzclaw Gen Chem Sg 9ed** Holtzclaw,1991 **Chemical Building Blocks** Michael J. Padilla,David V. Frank,John G. Little,Steve Miller,2000 Organic Chemistry Study Guide & Problems Book Brent L. Iverson,1998 Companion to Organic chemistry 2nd ed by William H Brown Provides a detailed section by section overview of the major points covered in the text All of the problems from the text are reprinted here with detailed stepwise solutions Shipman Phys Sci 6e Study Guide James T. Shipman,Jerry D. Wilson,1990 *Focus on Physical Science California Edition* Michael J. Padilla,2007 Study Guide: Sg Concepts in Biology Eldon D.. Enger,Enger,Frederick C.. Ross,2003 *Study Guide for Atoms, Molecules, and Life* Michael S. Matta,1981 *An Introductory Guide to EC Competition Law and Practice* Valentine Korah,1994 *Prentice Hall Science Explorer* Michael J. Padilla,2002 **Chemical Interactions** ,2005 **Basic Concepts of Chemistry, Study Guide** Leo J. Malone,2000 The text's three main goals are to introduce chemistry as a living relevant science to encourage learning and critical thinking and to help readers overcome the math difficulties that impede their progress in chemistry Designed to help readers master the principles of general chemistry As a prep book it promotes active involvement with the material There are special features throughout that reinforce concepts and help to develop strong problem solving and study skills Updated to Include an Interactive Learning Ware problems CD containing several of the chapter ending problems from the book in an interactive tutorial with feedback to help readers set up and solve problems **Resources in Education** ,1982 **New York Review of the Telegraph and Telephone and Electrical Journal** ,1905

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