

Study Guide Unit 9

1. KNOW the following vocabulary words (this means you NEED to be able to DEFINE them ALL):

- a. Solution- m. Solute-
- b. Homogeneous- n. Solvent-
- c. Heterogeneous-
- d. Soluble- o. Acid-
- e. Insoluble- p. Base-
- f. Miscible-
- g. Immiscible-
- h. Solubility- q. Titration-
- i. Alloy- r. Analyte-
- j. Saturated- s. Titrant-
- k. Unsaturated-
- l. Supersaturated-
- 2. How is the solubility of a gas affected by temp. and pressure?
Pressure up = solubility up
Temp up = solubility down
- 3. Under what conditions are salts most soluble?
High temperatures

4. Use the solubility curve to the right to answer the following.

a. What mass of KNO_3 will make a saturated solution at 60°C ? **100g**

b. Which of the compounds shown on the solubility curve decrease solubility as temperature increases?

NH_3 , $\text{Ca}_2(\text{SO}_4)_3$

c. What type of solution do I have if I **add** 110 g of NaNO_3 in 100g of 20°C water?

Saturated

d. What mass of KCl can go into solution at 80°C ?

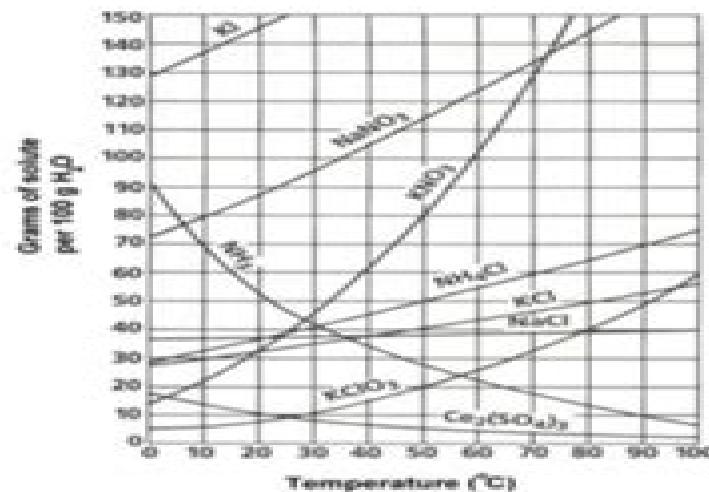
50 g

e. What is the least soluble compound at 20°C ?

KClO_3

f. Excluding KI , what is the most soluble compound at 80°C ?

KNO_3



Study Guide Solutions Answer Key

Irvin N. Gleim

Study Guide Solutions Answer Key:

Chemical Principles Study Guide/Solutions Manual John Krenos,Joseph Potenza,Carl Hoeger,2007-01-18 Written for general chemistry courses Chemical Principles helps students develop chemical insight by showing the connection between chemical principles and their applications *Study Guide and Solutions Manual* Bruce A. Chase,Peter J. Russell,2005-06 This student resource contains chapter outlines of text material solutions to all end of chapter problems key terms suggestions for analytical approaches problem solving strategies and a variety of additional questions for student practice Also featured are questions that relate to chapter specific animations and iActivities Student Study Guide/Solutions Manual for Essentials of General, Organic, and Biochemistry Denise Guinn,Rebecca Brewer,Rachel C. Lum,2009-09-15 The Student Study Guide and Solutions Manual provides students with a combined manual designed to help them avoid common mistakes and understand key concepts After a brief review of each section s critical ideas students are taken through stepped out worked examples try it yourself examples and chapter quizzes all structured to reinforce chapter objectives and build problem solving techniques The solutions manual includes detailed solutions to all odd numbered exercises in the text

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