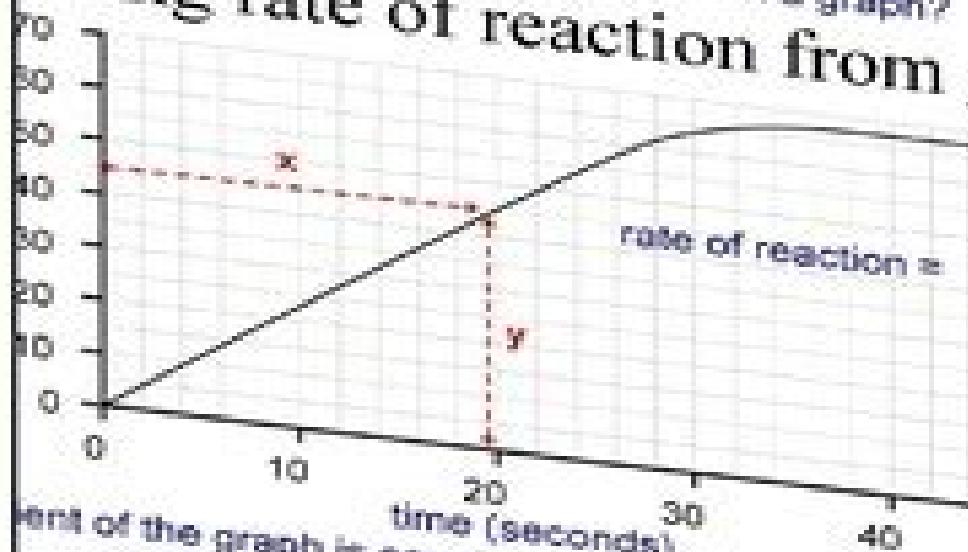


# TERMOCHIMISTRAV

How can the rate of reaction be calculated from a graph?



The gradient of the graph is equal to the initial rate of reaction.

$$\text{reaction rate} = \frac{45 \text{ cm}^3}{20 \text{ s}}$$

rate of reaction =

## ENDO- AND EXO- REACTIONS

In an exothermic reaction, as products form, it speeds up. Once there will come a time when the forward and backward reactions are equal and opposite; the reaction has



flow into or out of the system (e.g. pressure, temperature, reactants)

reaction at constant temperature

Exothermic reaction  
heat released by the system  
less than zero

Endothermic reaction  
heat absorbed by the system  
more than zero

# REACTIONS

# Thermochemistry Core Teaching Resources

**Huangqi Zhang**

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*Open educational resources: policy, costs, transformation* Miao, Fengchun, Mishra, Sanjaya, McGreal, Rory, 2016-04-18

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### **Teaching School**

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### **Biomedical Index to PHS-supported Research** , 1991

**Basic Thermochemistry in Materials Processing** Gabriel Plascencia, David Jaramillo, 2017-02-28 This book provides the reader with some thermochemistry notes The intention is to provide a simple easy to understand text which serves as a complimentary material to more complex books It also provides students and those beginning in the field with several application examples used in different areas of materials processing The book presents fully solved problems some quite often found in major metallurgical operations

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