

Subatomic Particles and Isotopes Practice

3. Complete the table below.

Element Name	Atomic Number	Mass Number	Number of protons	Number of neutrons	Number of electrons	Isotopic Mass
Argon		37				
	18	38				
	19			38		
Neon		20				
				20	10	
				20	10	
Sulfur				38		
				39	18	

4. Answer the following questions:

a) An atom contains 95 protons. What is the element symbol?

b) An atom contains 30 protons, 39 neutrons, and 30 electrons. Identify the mass number of this atom. _____

c) What is the atomic number of bromine? _____

d) How many neutrons in an atom of Mg-24? _____

e) What is the atomic number of Br-82? _____

f) An atom contains 30 protons. What is the element symbol?

g) An atom contains 37 protons, 38 neutrons, and 37 electrons. Identify the mass number of this atom. _____

Subatomic Particle Gizmo Answers Key

Aidan Smith

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