

Name \_\_\_\_\_

Date \_\_\_\_\_

**Subatomic Particles and Isotopes Practice**

1. Complete the table below.

Element Name	Atomic Number	Mass Number	Number of protons	Number of neutrons	Number of electrons	Isotope Notation
Oxygen		17				
	15	31				
	39			50		
Neon		20				
				10	9	
			79	128		
Iron				30		
				35	29	

2. Answer the following questions:

a) An atom contains 55 protons. What is the element symbol?

\_\_\_\_\_

b) An atom contains 31 protons, 39 neutrons, and 31 electrons. Identify the mass number of this atom. \_\_\_\_\_

c) What is the atomic number of bromine? \_\_\_\_\_

d) How many neutrons in an atom of Hg-201? \_\_\_\_\_

e) What is the atomic number of Zn-65? \_\_\_\_\_

f) An atom contains 33 protons. What is the element symbol?

\_\_\_\_\_

g) An atom contains 17 protons, 18 neutrons, and 17 electrons. Identify the mass number of this atom. \_\_\_\_\_

# Subatomic Particle Gizmo Answers Key

**Aidan Smith**

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

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