

Name: _____

Unit 1 Section B: Counting Subatomic Particles Worksheet

Using a periodic table, fill in the chart with the correct information. Assume that all atoms are electrically neutral.

	Name	Symbol	Atomic Number	Mass Number	# of Protons	# of Neutrons	# of Electrons
1.	boron	B	5	11	5	6	5
2.	zinc					31	
3.		K		42			
4.			22			27	
5.				122			51
6.		U				146	
7.					47	60	
8.	fermium			257			
9.	cesium					79	55
10.			42	97			
11.				120	78		
12.	krypton					47	
13.			86	222			
14.		N		15			
15.					7	7	

1. What similarities exist between the answers to questions 14 and 15 on the table?

2. What differences exist between the answers to questions 14 and 15 on the table?

3. What term can be applied to atoms like those described in questions 14 and 15 on the table?

4. What is the difference between mass number (used in this exercise) and atomic weight (found on the periodic table)?

Subatomic Particles Gizmo Answer Key

Randy L. Bell

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