

Mole Conversions Practice Answer Key

Molar Mass – Calculate the Molar Mass for the following.

1. What is the molar mass of Magnesium metal?

$$24.305 \text{ g/mol}$$

2. Calculate the molar mass of NaCl

$$\text{Na: } 22.990 \times 1 = 22.990$$

$$\text{Cl: } 35.453 \times 1 = 35.453 \rightarrow 58.443 \text{ g/mol}$$

3. What is the molar mass of NaOH?

$$\text{Na: } 1 \times 22.990 = 22.990$$

$$\text{O: } 1 \times 15.999 = 15.999$$

$$\text{H: } 1 \times 1.008 = 1.008 \rightarrow 39.997 \text{ g/mol}$$

Conversions – Calculate the following the conversions. Use your Mole Map to help you!

1. Determine the number of atoms in 0.50 moles He.

$$0.5 \text{ mol He} \times \frac{6.02 \times 10^{23} \text{ atoms He}}{1 \text{ mol He}} = 3.01 \times 10^{23} \text{ atoms He}$$

2. Determine the number of moles in 80.00 g O₂

$$0.2 \times 15.999 = 31.998 \text{ g/mol}$$

$$\rightarrow 1 \text{ mol} = 31.998 \text{ g}$$

$$80.00 \text{ g O}_2 \times \frac{1 \text{ mol O}_2}{31.998 \text{ g O}_2} = 2.500 \text{ mol O}_2$$

3. Determine the volume of 0.24 mol of NO₂.

$$0.24 \text{ mol NO}_2 \times \frac{22.4 \text{ L NO}_2}{1 \text{ mol NO}_2} = 5.376 \text{ L NO}_2$$

4. Determine the number of moles in 7.5×10^{23} formula units of KNO₃

$$7.5 \times 10^{23} \text{ f.u. KNO}_3 \times \frac{1 \text{ mol KNO}_3}{6.02 \times 10^{23} \text{ f.u. KNO}_3} = 1.2458 \times 10^{-4} \text{ mol KNO}_3$$

5. Determine the number of moles in 202.22g KNO₃

$$\text{K: } 1 \times 39.098 = 39.098$$

$$\text{N: } 1 \times 14.007 = 14.007$$

$$\text{O: } 3 \times 15.999 = 47.997 \rightarrow 101.102 \text{ g/mol}$$

6. Determine the number of moles in 32.5L of N₂ at STP

$$32.5 \text{ L N}_2 \times \frac{1 \text{ mol N}_2}{22.4 \text{ L N}_2} = 1.451 \text{ mol N}_2$$

$$202.22 \text{ g KNO}_3 \times \frac{1 \text{ mol KNO}_3}{101.102 \text{ g KNO}_3} = 2.00 \text{ mol KNO}_3$$

7. Determine the number of moles in 3.4×10^{24} formula units of NaCl.

$$3.4 \times 10^{24} \text{ f.u. NaCl} \times \frac{1 \text{ mol NaCl}}{6.02 \times 10^{23} \text{ f.u. NaCl}} = 564.784 \text{ mol NaCl}$$

8. Determine the moles in 36.5L of carbon monoxide

$$36.5 \text{ L CO} \times \frac{1 \text{ mol CO}}{22.4 \text{ L CO}} = 1.629 \text{ mol CO}$$

Skill Practice 30 Mole Conversion Practice Answers

Xiang Xie

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der klimawandel die klimakatastrophe sind in aller munde vielfach reden wir vom klima meinen aber eigentlich das wetter wie sich die idee vom klima selbst über die zeiten

gesundheitliche folgen des klimawandels bzga - Sep 25 2023

der klimawandel beeinflusst die menschliche gesundheit in vielfältiger weise extreme wetter und umweltveränderungen können bestehende gesundheitsprobleme verstärken und neue verursachen um unsere gesundheit hier und jetzt zu schützen müssen wir wissen was auf