

# Molarity Practice

1. Calculate the molarity of the following solutions:
  - a. 10.0 g of NaCl in 250 mL of solution
  - b. 50.0 g of sucrose, C<sub>12</sub>H<sub>22</sub>O<sub>11</sub>, in 250 mL of solution
  - c. 10.0 g of NaCl in 250 mL of solution
2. Calculate the molarity of the following solutions:
  - a. 10.0 g of NaCl in 250 mL of solution
  - b. 50.0 g of sucrose, C<sub>12</sub>H<sub>22</sub>O<sub>11</sub>, in 250 mL of solution
  - c. 10.0 g of NaCl in 250 mL of solution
3. Calculate the molarity of the following solutions for the following concentrations:
  - a. 10.0 g of NaCl in 250 mL of solution
  - b. 50.0 g of sucrose, C<sub>12</sub>H<sub>22</sub>O<sub>11</sub>, in 250 mL of solution
4. Calculate the molarity of the following solutions for the following concentrations:
  - a. 10.0 g of NaCl in 250 mL of solution
  - b. 50.0 g of sucrose, C<sub>12</sub>H<sub>22</sub>O<sub>11</sub>, in 250 mL of solution
5. Calculate the molarity of the following solutions for the following concentrations:
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## Skill Practice 53 Answers Chemistry

**IM Harris**

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