

# Single Replacement Reaction

A single replacement reaction occurs when one element replaces another element in a compound.



*Examples:*



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A single replacement reaction or single displacement reaction is when one element replaces another in a compound.

Get the definition of a single replacement reaction or single displacement reaction. Get examples of single replacement reactions and learn how to use the [metal reactivity series](#) to predict whether a reaction will occur and the products.

## Single Replacement Reaction Definition

A **single replacement reaction** is a chemical reaction where one element replaces another in a compound. It is also known as a **single displacement reaction**. The general form of a single replacement reaction chemical equation is:



Single replacement reactions occur when A is more reactive than B or [product](#) AC is more stable than BC. A and B can be either two metals (including hydrogen; C is [an anion](#)) or else two halogens (C is a cation). If BC and AC are in aqueous solutions, C acts as a spectator ion.

## Single Replacement Reaction Examples

There are two different scenarios for single replacement reactions. In one form of the reaction, one cation replaces the other. In the other form of the reaction, one anion replaces the other.

### Cation Replacement Examples

Usually the cation is a metal, but it doesn't have to be. Here are examples of single replacement reactions involving the cations:

# Single Replacement Reactions Study Guide

**Leo J. Malone**



## Single Replacement Reactions Study Guide:

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general chemistry As a prep book it promotes active involvement with the material There are special features throughout that reinforce concepts and help to develop strong problem solving and study skills Updated to Include an Interactive Learning Ware problems CD containing several of the chapter ending problems from the book in an interactive tutorial with feedback to help readers set up and solve problems

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