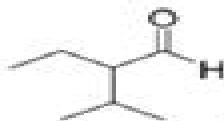


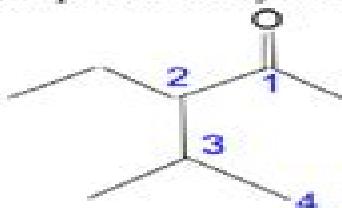
1. What is the correct IUPAC name of the following compound?



- a. 2-isopropylbutanal
- b. 2-ethyl-3-methylbutanal
- c. 2-isopropylbutanone
- d. 3-ethyl-2-methylbutanal

Correct Answer: B

Explanation: When naming aldehydes, you always want to change the -e from the parent name to -al. For example, hexane will be switched to hexanal if an aldehyde is present. We can already eliminate c, because we are not dealing with any ketones as our functional group. You always want to start where the carbonyl double bond is, so



that would be 1.

Now that the compound is numbered look at the answer choices. A cannot be the answer because there is no isopropyl group on carbon 2. Also, it's missing the other substituent of carbon 3. Answer choice B, makes the most since, because answer choice D would mean that the carbonyl double would not be given the smallest number. You can never assign the larger number to the aldehyde, ketone, double bond, or triple bond; always try to give it the smallest number.

2. Rank the compounds in order of decreasing reactivity with LiAlH4

Siva Group Organic Chemistry 2 Answer

Yijin Wang

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1. Understanding the eBook Siva Group Organic Chemistry 2 Answer
 - The Rise of Digital Reading Siva Group Organic Chemistry 2 Answer
 - Advantages of eBooks Over Traditional Books
2. Identifying Siva Group Organic Chemistry 2 Answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Siva Group Organic Chemistry 2 Answer
 - User-Friendly Interface
4. Exploring eBook Recommendations from Siva Group Organic Chemistry 2 Answer
 - Personalized Recommendations
 - Siva Group Organic Chemistry 2 Answer User Reviews and Ratings
 - Siva Group Organic Chemistry 2 Answer and Bestseller Lists
5. Accessing Siva Group Organic Chemistry 2 Answer Free and Paid eBooks
 - Siva Group Organic Chemistry 2 Answer Public Domain eBooks
 - Siva Group Organic Chemistry 2 Answer eBook Subscription Services
 - Siva Group Organic Chemistry 2 Answer Budget-Friendly Options
6. Navigating Siva Group Organic Chemistry 2 Answer eBook Formats

- ePub, PDF, MOBI, and More
 - Siva Group Organic Chemistry 2 Answer Compatibility with Devices
 - Siva Group Organic Chemistry 2 Answer Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Siva Group Organic Chemistry 2 Answer
 - Highlighting and Note-Taking Siva Group Organic Chemistry 2 Answer
 - Interactive Elements Siva Group Organic Chemistry 2 Answer
8. Staying Engaged with Siva Group Organic Chemistry 2 Answer
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Siva Group Organic Chemistry 2 Answer
9. Balancing eBooks and Physical Books Siva Group Organic Chemistry 2 Answer
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Siva Group Organic Chemistry 2 Answer
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Siva Group Organic Chemistry 2 Answer
- Setting Reading Goals Siva Group Organic Chemistry 2 Answer
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Siva Group Organic Chemistry 2 Answer
- Fact-Checking eBook Content of Siva Group Organic Chemistry 2 Answer
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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