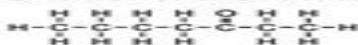


## Quiz & Worksheet - Structural Formula

1. Which of the following shows the correct condensed structural formula of a compound that has five carbon (C) atoms, 12 hydrogen (H) atoms and one oxygen (O) atom?

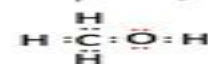
- ☐  $\text{CH}_3 - \underset{\text{CH}_2}{\overset{\text{CH}_2}{\text{C}}} - \text{CH}_2 - \overset{\text{O}}{\parallel}{\text{C}} - \text{H}$
- ☐  $\text{CH}_3 - \underset{\text{CH}_2}{\overset{\text{CH}_3}{\text{C}}} - \text{CH}_2 - \overset{\text{O}}{\parallel}{\text{C}} - \text{H}$
- ☐  $\text{CH}_3 - \underset{\text{CH}_3}{\overset{\text{CH}_3}{\text{C}}} - \text{CH} - \text{CH}_2 - \text{O}$
- ☐  $\text{CH}_3 - \underset{\text{CH}_3}{\overset{\text{CH}_3}{\text{C}}} - \text{CH}_2 - \overset{\text{O}}{\parallel}{\text{C}} - \text{H}$

2. Which of the following is the correct skeletal structural formula for the following compound?



- ☐
- ☐
- ☐
- ☐

3. What is the correct line-bond structural formula for the compound shown below? (The black dots are bonding electrons between the atoms and the red dots are nonbonding electrons.) Hint: Since this is an organic compound, nonbonding electrons are, in general, not shown in the line-bond structural formula



- ☐  $\text{H} - \underset{\text{H}}{\overset{\text{H}}{\text{C}}} - \overset{\text{O}}{\parallel}{\text{C}} - \text{H}$
- ☐  $\text{H} - \underset{\text{H}}{\overset{\text{H}}{\text{C}}} - \text{O} - \text{H}$
- ☐
- ☐  $\text{CH}_3 - \text{O} - \text{H}$

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# Skeletal Formula Practice Problems

**John McMurry**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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**Organic Chemistry** T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2022-05-26 Organic Chemistry 13th Edition continues Solomons Fryhle and Snyder s tradition of excellence in teaching and preparing students for success in both the classroom and beyond Central to the authors is their approach in emphasizing organic chemistry s relationship between structure and reactivity To accomplish this the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms The authors philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters The structural aspects of the authors approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us

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This book delves into Skeletal Formula Practice Problems. Skeletal Formula Practice Problems is a crucial topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Skeletal Formula Practice Problems, encompassing both the fundamentals and more intricate discussions.

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- Chapter 1: Introduction to Skeletal Formula Practice Problems
- Chapter 2: Essential Elements of Skeletal Formula Practice Problems
- Chapter 3: Skeletal Formula Practice Problems in Everyday Life
- Chapter 4: Skeletal Formula Practice Problems in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Skeletal Formula Practice Problems. This chapter will explore what Skeletal Formula Practice Problems is, why Skeletal Formula Practice Problems is vital, and how to effectively learn about Skeletal Formula Practice Problems.
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This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Skeletal Formula Practice Problems.

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