

SECTION A – Multiple-choice questions

Instructions for Section A

Answer **all** questions.
Choose the response that is **correct** or **best** answers the question.
A correct answer scores 1, an incorrect answer scores 0.
No mark will be given if more than one answer is completed for any question.
Marks will **not** be deducted for incorrect answers.

Question 1

The skeletal structure of a molecule is shown below.



The molecule is most likely

- A. to be a significant component of LPG.
- B. to be a component of petrol.
- C. an example of biodiesel.
- D. a typical component of petrodiesel.

Question 2

The density of a sample of petrodiesel is 0.85 g mL^{-1} . The amount of energy released from the complete combustion of 1.0 L of diesel will be, in kJ,

- A. 38
- B. 1040
- C. 38000
- D. 45000

Question 3

Which equation is important in the production of a biofuel?

- A. $\text{C}_3\text{H}_8(\text{g}) + 3.5\text{O}_2(\text{g}) \rightarrow 2\text{CO}_2(\text{g}) + 3\text{H}_2\text{O}(\text{l})$
- B. $\text{C}_2\text{H}_6(\text{g}) + 2.5\text{O}_2(\text{g}) \rightarrow 2\text{CO}(\text{g}) + 3\text{H}_2\text{O}(\text{l})$
- C. $\text{C}_6\text{H}_{12}\text{O}_6(\text{aq}) + 6\text{O}_2(\text{g}) \rightarrow 6\text{CO}_2(\text{g}) + 6\text{H}_2\text{O}(\text{aq})$
- D. $\text{C}_6\text{H}_{12}\text{O}_6(\text{aq}) \rightarrow 2\text{C}_2\text{H}_5\text{O}(\text{aq}) + 2\text{CO}_2(\text{g})$

Question 4

The complete combustion of 0.025 mol of a fuel produces 0.10 mol of CO_2 . The fuel is most likely to be

- A. methane.
- B. ethane.
- C. butane.
- D. octane.

Unit 3 Chemistry Trial Exams

Andrew Shortell, Judith Stillman



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