

SECTION A – Multiple-choice questions

Instructions for Section A

Answer all questions.

Choose the response that is correct or best answers the question.

A correct answer scores 1, an incorrect answer scores 0.

No mark will be given if more than one answer is completed for any question.

Marks will not be deducted for incorrect answers.

Question 1

The skeletal structure of a molecule is shown below.



The molecule is most likely

- A. to be a significant component of LPG.
- B. to be a component of petrol.
- C. an example of biodiesel.
- D. a typical component of petrodiesel.

Question 2

The density of a sample of petrodiesel is 0.85 g mL^{-1} . The amount of energy released from the complete combustion of 1.0 L of diesel will be, in kJ,

- A. 38
- B. 1040
- C. 38000
- D. 45000

Question 3

Which equation is important in the production of a biofuel?

- A. $\text{C}_2\text{H}_6(\text{g}) + 3.5\text{O}_2(\text{g}) \rightarrow 2\text{CO}_2(\text{g}) + 3\text{H}_2\text{O}(\text{l})$
- B. $\text{C}_2\text{H}_6(\text{g}) + 2.5\text{O}_2(\text{g}) \rightarrow 2\text{CO}(\text{g}) + 3\text{H}_2\text{O}(\text{l})$
- C. $\text{C}_6\text{H}_{12}\text{O}_6(\text{aq}) + 6\text{O}_2(\text{g}) \rightarrow 6\text{CO}_2(\text{g}) + 6\text{H}_2\text{O}(\text{aq})$
- D. $\text{C}_6\text{H}_{12}\text{O}_6(\text{aq}) \rightarrow 2\text{C}_2\text{H}_6\text{O}(\text{aq}) + 2\text{CO}_2(\text{g})$

Question 4

The complete combustion of 0.025 mol of a fuel produces 0.10 mol of CO_2 . The fuel is most likely to be

- A. methane.
- B. ethane.
- C. butane.
- D. octane.

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