

Biochemistry

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Chapter 1: Overview

- 1. Biochemistry has unity because of biological evolution**
- 2. Biology uses mainly six elements plus inorganic salts**
 - 65% Oxygen, 18% Carbon, 10% Hydrogen, 3% Nitrogen, 2% Calcium (salt).
 - 1.1% Phosphorus, 0.35% Potassium
 - Small amounts of S, Na, Cl, Mg, Fe, I
 - Water (80-80% of our mass). It participates in reactions and is a limited solvent
- 3. Biochemistry consists of five classes of organic compounds**
 1. Carbohydrates
 2. Lipids
 3. Nucleic Acids
 4. Proteins
 5. Small organic molecule, inorganic molecules & ions
- 4. DNA provides an example of how structure determines function**
 - DNA is very polar
- 5. Important biochemical interactions include electrostatic interactions, hydrogen bonds, and hydrophobic interactions**
 - Electrostatic attractions
 - Hydrogen bonds: involve very strong partial charges and stronger than the average polar bond
 - Hydrophobic interactions (london dispersion/ van der Waals): most important because most abundant. A water "poor" area created by hydrophobic interactions
- 6. Important types of reactions in biochemistry include redox and acid-base reactions**
 - a. Redox reactions
 - b. Acid-base reactions
 - Bronsted-Lowry:
 - acids = proton donor
 - bases = proton acceptor
 - **Moiety:** an atom or group of atoms that composes a molecule or part of a molecule. (pK_a is the acidity or basicity of a moiety)
 - A moiety is protonated when the pH is lower than its pK_a , and unprotonated when the pH is higher than its pK_a
- 7. Modern biochemistry includes discovery-based as well as hypothesis-driven methods**
 - Discovery based: genetic screening (what gene caused the mutation), biological surveys (what living organisms are in this sector of land)

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The Senior Laboratory Technician Biochemistry Passbook R prepares you for your test by allowing you to take practice exams in the subjects you need to study It provides hundreds of questions and answers in the areas that will likely be covered on your upcoming exam including but not limited to laboratory principles and practices use care and operation of laboratory instruments record keeping supervision

biochemistry and more **Clinical Medicine; Examination Review** George Horemis,1972 *Biochemistry Basics* Milin Kurup,2020-12-21 The Biochemistry Basics Biochemistry and Molecular Biology Study Guide was created by a renowned student from the University of Florida and includes all notes diagrams and study guides for all the important subjects covered in Biochemistry Molecular Biology Genetics and Microbiology Milin Kurup is a double major in B S Microbiology and

Cognitive and Behavioral Neuroscience student from the University of Florida In addition to his degree Milin is a UF Biochemistry BCH4024 Study Instructor Group Leader a Microbiology MCB3020L Teaching Assistant a Genetics PCB4522 Teaching Assistant and a Neuroscience Research Assistant at the University of Florida While many of these classes cover high density material this study guide hopes to organize and condense the whole curriculum into short page review sheets In

the author s time of instruction and study he organized a collection of all reactions mechanisms processes and concepts all studied in Biochemistry Genetics and Microbiology Overall this biochemistry study guide covers topics such as biomolecule structures Protein Carbohydrate Nucleic Acids and Lipids biomolecules function biomolecule metabolism Protein Metabolism Carbohydrate Metabolism Nucleic Acid Metabolism and Lipid Metabolism physiological biochemical relationships genetics and biological microbiological biochemical processes Overall the guide is organized into 1 3 page summaries of each specific topic and acts as a study guide for those who hope to study individual concepts in detail All sections include detailed

diagrams color coded notes labeled illustration and detailed descriptions for effective comprehension In addition to class studies many students also have used this study guide as an MCAT review guide The short and condensed review pages have helped many student organize and categorize important topics as they continue to study for the MCAT Ultimately this

organized set can be extremely useful for students review especially before class exams school projects standardized test and much more

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