

## Study Guide for Chemistry EOC Assessment

Your test will consist of multiple-choice questions over the following topics.

1. Metric units for mass, length, volume
2. Metric to metric conversions (K H Da m/l/g D C M)
3. Know how to put numbers in scientific notation and take them out.
4. Significant Digit rules
5. Difference between chemical and physical properties and changes. There will be multiple questions on this.
6. Ways to tell if a chemical change (chemical reaction) is taking place.
7. Pure substances and mixtures
  - a. heterogeneous mixture
  - b. homogeneous mixture
8. Density & how it relates to floating and sinking
9. Use the equation for density 
$$\text{Density} = \frac{\text{Mass}}{\text{Volume}}$$
$$(\text{g/ml}) \quad (\text{g}) \quad (\text{ml})$$
10. Know the phases of matter and the characteristics of each phase.
11. Law of Conservation of Matter
12. What is a mole in chemistry?
13. Convert grams to moles, and moles to grams
14. Define an atom
15. Structure of an atom and where each subatomic particle is located.
  - a. Nucleus (contains protons and neutrons)
  - b. Protons (positive)
  - c. Neutrons (neutral)
  - d. Electrons (negative)
16. Isotopes
17. Elements
18. If given atomic # and mass, find the number of protons and neutrons in an atom.
19. Find the number of valence electrons, and tell what it says about reactivity.
20. Periodic Table
  - a. Know about the families
  - b. Group vs. period
21. Know about the properties of metals and nonmetals.
  - a. Ductile, malleable, conductors
22. Ionic & Covalent Compounds
  - a. How do they form? (Transfer or gain electrons)
  - b. Characteristics of Each
23. Be able to name ionic compounds
24. Predict oxidation numbers
25. Chemical Equations:
  - a. Know the parts of an equation
  - b. 5 main types of reactions-know about them and be able to identify them (Synthesis, Decomposition, Single Displacement, Double Displacement, Combustion)
26. Acids & Bases:
  - a. Macro and Microscopic properties of each
  - b. What ions do they form, how do they affect litmus, what is their pH range?
  - c. Identify examples of each- Mono, di, and polyprotic acids
  - d. Tell if they are strong or weak
  - e. Acid-base indicator

# Study Guide For Chemistry Eoc

**John E. McMurry, Robert C. Fay, Jill  
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Zubricky**

## **Study Guide For Chemistry Eoc:**

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**Roadmap to the Virginia SOL** Princeton Review,2005 Roadmap to the Virginia SOL EOC Chemistry includes strategies that are proven to enhance student performance The experts at The Princeton Review provide content review of the crucial material most likely to appear on the test detailed lessons complete with test taking techniques for improving test scores 2 complete practice Virginia SOL EOC Chemistry tests

*Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today* Seager/Slabaugh,2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the

learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application **STAAR EOC Chemistry Assessment Secrets Study Guide** Mometrix Media LLC,2016-10-24 Includes Practice Test Questions STAAR EOC Chemistry Assessment Secrets helps you ace the State of Texas Assessments of Academic Readiness without weeks and months of endless studying Our comprehensive STAAR EOC Chemistry Assessment Secrets study guide is written by our exam experts who painstakingly researched every topic and concept that you need to know to ace your test Our original research reveals specific weaknesses that you can exploit to increase your exam score more than you ve ever imagined STAAR EOC Chemistry Assessment Secrets includes The 5 Secret Keys to STAAR Success Time is Your Greatest Enemy Guessing is Not Guesswork Practice Smarter Not Harder Prepare Don t Procrastinate Test Yourself A comprehensive General Strategy review including Make Predictions Answer the Question Benchmark Valid Information Avoid Fact Traps Milk the Question The Trap of Familiarity Eliminate Answers Tough Questions Brainstorm Read Carefully Face Value Prefixes Hedge Phrases Switchback Words New Information Time Management Contextual Clues Don t Panic Pace Yourself Answer Selection Check Your Work Beware of Directly Quoted Answers Slang Extreme Statements Answer Choice Families Along with a complete in depth study guide for your specific STAAR exam and much more [STAAR Success Strategies Eoc Chemistry: STAAR Test Review for the State of Texas Assessments of Academic Readiness](#) Mometrix Media LLC,Staar Exam Secrets Test Prep,2014-03-31 STAAR Success Strategies EOC Chemistry helps you ace the State of Texas Assessments of Academic Readiness without weeks and months of endless studying Our comprehensive STAAR Success Strategies EOC Chemistry study guide is written by our exam experts who painstakingly researched every topic and concept that you need to know to ace your test Our original research reveals specific weaknesses that you can exploit to increase your exam score more than you ve ever imagined STAAR Success Strategies EOC Chemistry includes The 5 Secret Keys to STAAR Success Time is Your Greatest Enemy Guessing is Not Guesswork Practice Smarter Not Harder Prepare Don t Procrastinate Test Yourself A comprehensive General Strategy review including Make Predictions Answer the Question Benchmark Valid Information Avoid Fact Traps Milk the Question The Trap of Familiarity Eliminate Answers Tough Questions Brainstorm Read Carefully Face Value Prefixes Hedge Phrases Switchback Words New Information Time Management Contextual Clues Don t Panic Pace Yourself Answer Selection Check Your Work Beware of Directly Quoted Answers Slang Extreme Statements Answer Choice Families Along with a complete in depth study guide for your specific STAAR exam and much more **Pamphlet - Dept. of the Army** United States. Department of the Army,1951 **Study Guide for Chemical Principles [by] Steven S. Zumdahl** Steven S. Zumdahl,Paul B. Kelter,1995 The Study Guide reflects the unique problem solving approach taken by the Chemical Principles text The new edition of the Study Guide includes many new worked out examples [Introductory Chemistry](#) Mike Lucarelli,David Alan Proctor,2002

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