

## Concentration Worksheet – Answers

- 1) How many grams of beryllium chloride are needed to make 125 mL of a 0.050 *M* solution?  
**0.50 grams**
- 2) How many grams of beryllium chloride would you need to add to 125 mL of water to make a 0.050 molal solution?  
**0.50 grams**
- 3) The density of ethanol is 0.789 g/mL. How many grams of ethanol should be mixed with 225 mL of water to make a 4.5% (v/v) mixture?  
**7.99 grams**
- 4) Explain how to make at least one liter of a 1.25 molal ammonium hydroxide solution.  
**Dissolve 43.8 grams (1.25 moles) of ammonium hydroxide in 1 L H<sub>2</sub>O.**
- 5) What is the molarity of a solution in which 0.45 grams of sodium nitrate are dissolved in 265 mL of solution.  
**0.020 *M***
- 6) What is the mole fraction of sulfuric acid in a solution made by adding 3.4 grams of sulfuric acid to 3,500 mL of water?  
 **$1.8 \times 10^{-4}$**
- 7) What will the volume of a 0.50 *M* solution be if it contains 25 grams of calcium hydroxide?  
**680 mL**
- 8) How many grams of ammonia are present in 5.0 L of a 0.050 *M* solution?  
**4.3 grams**

# Study Guide For Solution Concentration Answer Key

**N. Sue Barnes**



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