

Concentration Worksheet – Answers

- 1) How many grams of beryllium chloride are needed to make 125 mL of a 0.050 *M* solution?
0.50 grams
- 2) How many grams of beryllium chloride would you need to add to 125 mL of water to make a 0.050 molal solution?
0.50 grams
- 3) The density of ethanol is 0.789 g/mL. How many grams of ethanol should be mixed with 225 mL of water to make a 4.5% (v/v) mixture?
7.99 grams
- 4) Explain how to make at least one liter of a 1.25 molal ammonium hydroxide solution.
Dissolve 43.8 grams (1.25 moles) of ammonium hydroxide in 1 L H₂O.
- 5) What is the molarity of a solution in which 0.45 grams of sodium nitrate are dissolved in 265 mL of solution.
0.020 *M*
- 6) What is the mole fraction of sulfuric acid in a solution made by adding 3.4 grams of sulfuric acid to 3,500 mL of water?
1.8 × 10⁻⁴
- 7) What will the volume of a 0.50 *M* solution be if it contains 25 grams of calcium hydroxide?
680 mL
- 8) How many grams of ammonia are present in 5.0 L of a 0.050 *M* solution?
4.3 grams

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N. Sue Barnes

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