

Unit 6 – Worksheet 4

Molecular Compounds

Properties

Basic structural unit

1. Name each of the following binary compounds of non-metallic elements
 - a. CBr_4
 - b. N_2P_3
 - c. PCl_3
 - d. ICl
 - e. N_2O
 - f. SiF_4
 - g. GeH_4
 - h. N_2Br_4
 - i. P_2S_5
 - j. SeO_2
 - k. NH_3
 - l. SiO_2
2. Write the formula for the following binary compounds of nonmetallic elements.
 - a. phosphorus trioxide
 - b. silicon tetrachloride
 - c. dinitrogen pentoxide
 - d. dinitrogen tetroxide
 - e. carbon monoxide
 - g. carbon dioxide
 - h. sulfur hexafluoride
 - i. dinitrogen tetrachloride
 - j. carbon tetrachloride
 - k. phosphorus pentafluoride
 - l. diphosphorus pentoxide

Unit 6 4 Molecular Compounds Answers

Henry Watts (F.C.S.)

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