

During this practical session, find the sample standard deviation below

Step 1: $K_1 = 100$ $K_2 = 1000$

1. Balance the weights
2. If possible, split 1.25 grams into 12.50 grams of K_1 (not to the 100mg accuracy)
3. After finding the amount of K_1 (which must be 12.50g) that is used as standard (standard weight)
4. Count 1.25 grams into balance of K_2
5. If this measurement is within accuracy is 1.25 grams of K_2 (which is the 12.50g)

Unit 7 Stoichiometry Study Guide

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